The Lomagundi-Jatuli carbon isotopic event recorded in the marble of the Tandilia System basement, Río de la Plata Craton, Argentina

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Abstract

The "Lomagundi-Jatuli event" corresponds to the most important δ^{13} C positive anomaly (\geq 5‰) globally reported in Palaeoproterozoic marine carbonates (between ~2.30 and 2.06 Ga). In the Tandilia System (Argentina), Río de la Plata Craton, this event was recorded in the basement marble of the San Miguel area. The calcite-diopside marble, hosted by biotite gneiss and intruded by 2.12 Ga garnet-leucogranite, was metamorphosed in amphibolite facies during the Transamazonian Cycle.

PAAS-normalised rare-earth elements (REE) and Y for the carbonate rocks are HREEenriched and display positive Eu and Y anomalies, typical of primary precipitates from a mixed hydrothermal-marine environment carbonate. Additionally, a truly negative Ce anomaly for all the samples indicates that the depositional environment was oxidising.

Positive δ^{13} C values ranging from +5.90 to +4.30‰ (V-PDB), and δ^{18} O from +17.45 to +13.84‰ (V-SMOW) were determined in this marble, both gradually decreasing towards the contact with the leucogranites. These values indicate that devolatilization reactions took place during the crystallisation of a wollastonite-vesuvianite-grossular-diopside skarn generated by the leucogranite intrusions into the marble. δ^{18} O values obtained from diopside and calcite crystals, in the marble sectors furthest from the contacts with leucogranite, allowed a 663°C to 623°C formation temperature to be calculated, considering oxygen in a calcite-diopside geothermometric pair. These temperatures are consistent with the metamorphic degree (amphibolite facies) reached in this portion of the basement.

Although the San Miguel marble shows petrographic and mineralogical evidence of regional and contact metamorphism, important geochemical and isotopic characteristics,

together with its estimated Palaeoproterozoic age, indicate that the marble protolith was a marine carbonate deposited during the "Lomagundi-Jatuli event".

Keywords: Lomagundi-Jatuli event; carbon isotope; metamorphism; Tandilia System;

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1. Introduction

The first stable isotope studies conducted on marine carbonates of the Lomagundi Group, Zimbabwe, which were deposited approximately in the 2.40-2.00 Ga interval, revealed an important positive anomaly in the δ^{13} C values (Schidlowski *et al.*, 1975, 1976). The anomaly was subsequently reported as much for well-preserved Palaeoproterozoic carbonates as others with superimposed metamorphism processes. This anomaly is of geological relevance, and was later named the "Lomagundi-Jatuli event" (LJE) by Melezhik *et al.* (2005).

Important advances in the knowledge of this event were accomplished in the Fennoscandian Shield (Fig. 1; Norway, Switzerland, and Finland) by Baker and Fallick (1989), Karhu (1993), and Melezhik and Fallick (1996); in Scotland by Baker and Fallick (1989); and in the North American Craton by Schidlowski *et al.* (1983), Mirota and Veizer (1994), Melezhik *et al.* (1997), and Bekker *et al.* (2003, 2006). Also, other outcrops from the South African cratons (Schidlowski *et al.*, 1976; Gauthier-Lafaye and Weber, 1989; Master *et al.*, 1990; Buick *et al.*, 1998, 2003; Bekker *et al.*, 2001; Schröder *et al.*, 2008; Master *et al.*, 2010; Martin *et al.*, 2013), the San Francisco Craton, Brazil (Bekker *et al.*, 2003; Maheshwari *et al.*, 2010), the Aravalli Craton, India (Maheshwari *et al.*, 1999, 2002, 2010), the north-eastern Sino-Korean Craton (Tang *et al.*, 2011), and from the Pilbara and Yilgarn Cratons, Australia (Lindsay and Brasier, 2002) have been studied (Fig. 1). Typically, these carbonates show δ^{13} C values between +5.00‰ and +16.00‰ or even higher (Lomagundi Group, Melezhik *et al.*, 2005). Although the beginning and duration of the "Lomagundi-Jatuli event" is still a matter of debate, records of stratigraphic and chemostratigraphic studies indicate that this event

started before 2.20 Ga, shortly after the Huronian glaciation ~2.45-2.22 Ga (Young *et al.*, 2001) and the Great Oxidation Event 2.45-2.32 Ga (Holland, 2002; Bekker *et al.*, 2003), and finished around 2.11-2.06 Ga (Bekker *et al.*, 2008), extending for more than 300 Ma (Karhu and Holland, 1996; Melezhik *et al.*, 2005, 2007). During this period, an increase in the organic C production consequently released to the oceans and atmosphere, accompanied by an increase in the O levels, produced C oxidation, generating important modifications both in the environment and in biological processes (Karhu and Holland, 1996). Thus, the characterisation of the space and time distribution of "Lomagundi-Jatuli event" carbonates is still an essential tool for the correlation and understanding of the processes that generated one of the most important modifications to the C cycle worldwide during the Palaeoproterozoic period.

This anomaly was also recognized by Maheshwari *et al.* (2010) in the Piedra Alta terrane, Uruguay, Río de la Plata Craton (Fig. 1; RPC). The carbonate is predominantly dolomitic, and lays in the San José Belt (Bossi *et al.*, 1993, 1998; Bossi and Cingolani, 2009), constituting a sedimentary succession that corresponds to the Paso Severino Formation. The unit has a depositional SHRIMP U-Pb age of 2.14 ± 0.21 Ga (obtained from meta-rhyolites from the top of this formation; Hartmann *et al.*, 2000a). Also, a minimum SHRIMP U-Pb age of 2.07 ± 0.06 Ga has been obtained from the Isla Mala granodiorite (Hartmann *et al.*, 2000a). The carbonate has a wide range of δ^{13} C values varying from -5.00 to +11.60‰. Although the positive δ^{13} C values are typical of the Palaeoproterozoic carbonates, the negative ones would be associated with remineralisation of the organic matter in early diagenetic environments (Maheshwari *et al.*, 2010).

In this work, we report δ^{13} C and δ^{18} O data, together with rare earth elements and Y analyses of basement marble of the San Miguel area, Tandilia System (Argentina), in order to stablish the marble origin and its relationship with the Lomagundi-Jatuli isotopic event.

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2. Geological setting

2.1 Río de la Plata Craton

The Río de la Plata Craton (RPC; Almeida *et al.*, 1973; Rapela *et al.*, 2007; Cingolani, 2010) is located in South America, and it is mainly exposed in Uruguay, Paraguay, south-eastern Brazil, and to some extent in Argentina (Fig. 2a). The Craton is composed of dominated Palaeoproterozoic (Rhyacian) igneous-metamorphic units (Cordani *et al.*, 2000; Rapela *et al.*, 2007), possible Archaean nuclei (Hartmann *et al.*, 2000b; Cingolani, 2010), and Mesoproterozoic belts and rocks (Oyhantçabal *et al.*, 2005; Gaucher *et al.*, 2011) mostly covered by Neoproterozoic to Lower Paleozoic sedimentary rocks, Cretaceous basalts from the Paraná basin, and by modern sediments restricted to some sectors.

The RPC was affected by two major cycles that correspond to the Transamazonian (or Transplatense, Santos *et al.*, 2017), 2.20-1.70 Ga, and the Brazilian, 0.80-0.50 Ga (Brito Neves *et al.*, 2014; Dalla Salda *et al.*, 1988; Pimentel and Fuck, 1992; Bossi *et al.*, 1993; Hartmann *et al.*, 2002a).

The westernmost edge of the RPC is in contact with the Pampia terrane in Argentina (Rapela *et al.*, 2007; Favetto *et al.*, 2015; Fig. 2a) and in Paraguay (Caapuçú or Tebycuary High; Fulfaro, 1996; Fig. 2a), while the eastern limit crops out in southern

Brazil, Encantadas-Tacuarembó domains (Hartmann *et al.*, 2007; Santos *et al.*, 2017), Curitiba and Luís Alves microcratons (Siga Júnior *et al.*, 2007), and in Uruguay in the Piedra Alta terrane (Santos *et al.*, 2003; Chernicoff *et al.*, 2014), Nico Pérez terrane (Gaucher *et al.*, 2009; Bossi and Cingolani, 2009; Santos *et al.*, 2017) and Rivera block (the last one is considered part of the Tacuarembó domain, Santos *et al.*, 2017) (Fig. 2b). The southernmost outcrops of the RPC are exposed in Argentina, at Martín García Island (Fig. 2a) and in the Tandilia region, (Fig. 2c; Dalla Salda *et al.*, 2005; Pankhurst *et al.*, 2003; Rapela *et al.*, 2007). However, some authors suggest, based on geochronological data, that the boundary of the RPC could be extended further south (to the limit with the Patagonia terrane; Ramos, 1988; Tohver *et al.*, 2012) and further north (Paranapanema domain, Santos *et al.*, 2017).

2.2 Tandilia System

The Tandilia System comprises a group of hills that reach 500 m above sea level, and are distributed along 350 km in a northwest-southeast direction. The set of rocks exposed in these hills corresponds predominantly to a Palaeoproterozoic basement covered by a sedimentary sequence of Lower Neoproterozoic to Silurian age (Fig. 2c). The basement, named the Buenos Aires Complex (Marchese and Di Paola, 1975), consists mainly of igneous and metamorphic rocks. The igneous rocks are acidic to intermediate granitoids with a great variability in their mineralogy and texture. They have a SHRIMP U-Pb age of 2.10-2.18 Ga (Chernicoff *et al.*, 2015; Cingolani *et al.*, 2002; Hartmann *et al.*, 2002a and b) and an Rb-Sr age between 2.15 and 1.77 Ga (Hart *et al.*, 1965; Halpern *et al.*, 1970; Linares and González, 1990; Pankhurst *et al.*, 2003). The metamorphic rocks correspond to granitic and tonalitic gneisses (with varied

characteristics), migmatites, amphibolites and scarce schists, marbles, and skarns. Mylonitic zones have also been described (Villar Fabre, 1954; Teruggi and Kilmurray, 1975; Cingolani and Dalla Salda, 2000) concentrated in two E-W trending belts: the Azul Megashear Zone that crops out along a 40 km stretch south of Azul city (Frisicale et al., 1999), and a 25-km-long belt located south of Tandil city (Dalla Salda, 1981). The most extensive outcropping metamorphic rocks of the region are located in the Balcarce area (Fig. 2c) and correspond to garnet gneisses, which have a SHRIMP U-Pb age of 2.17-2.07 Ga (Cingolani et al., 2002; Hartmann et al., 2002a). These rocks have variable fabrics that ranges from strongly foliated to massive or isotropic (Dalla Salda et al., 2005). The mineralogy includes plagioclase, potassium feldspar, quartz, garnets and/or biotite, and eventually pyroxene; also, some varieties of these gneisses comprise muscovite and sillimanite (Dalla Salda et al., 2005). These rocks exhibit evidence of partial melting, such as quartz-feldspathic concordant and discordant small bodies, and/or sectors enriched with mafites, garnet, and epidote (Dalla Salda et al., 2005). Migmatites are frequently found, however, they are difficult to identify since they are closely related to with gneisses and granitic rocks. They are made up of quartz-feldspar leucosomes, sometimes with neoformed garnets, and have Sm-Nd ages of 2.14 ± 0.88 Ga (Pankhurst *et al.*, 2003). Structures are diverse and vary from stromatic to nebulitic (Dalla Salda et al., 2005). Amphibolites occur as isolated bodies or xenoliths, included within the gneisses or the granitic plutons, in the central portion of the Tandilia System. They are composed of plagioclases, quartz, amphibole, epidote, and biotite (Quartino and Villar Fabre, 1967; Delpino, 2000; Dalla Salda et al., 2005). Schists are scarce, mainly micaceous, and they are found in the Balcarce and Azul city areas (Teruggi and Kilmurray, 1980).

Two types of marbles have been identified: dolomite marble at Punta Tota, Balcarce city (Delpino and Dristas, 2008), and calcite marble at San Miguel, near Tandil city (Quartino and Villar Fabre, 1967; Lajoinie *et al.*, 2013, 2014b; Lajoinie, 2015) (Fig. 2c). Both marbles show evidence of metasomatism, with further formation of skarns. Geothermobarometric data, obtained by Delpino and Dristas (2008) from dolomites and associated calc-silicate rocks of Punta Tota area, indicate metamorphic conditions of 750-800°C and 5-6kbar, followed by an isobaric cooling to 450-500°C.

A set of low-grade metamorphic rocks (low greenschist facies) which consists of metacherts, metagreywackes, and metabasalts (El Cortijo Formation, Teruggi *et al.*, 1988; Lajoinie *et al.*, 2017) crops out in the SW area of Tandil city (Fig. 2c). These rocks have been interpreted as the remaining portion of the oceanic crust linked to an island arc accreted to the Tandilia terrane during the Transamazonian Cycle (Teruggi *et al.*, 1988).

A Neoproterozoic sedimentary cover, composed of a calcareous-siliciclastic succession, overlays this basement (Fig. 2c). It comprises the Sierras Bayas Group (Dalla Salda and Iñiguez, 1979; Poiré *et al.*, 1984; Poiré, 1993) and the La Providencia Group, with ages of ~0.90-0.80 Ga to 0.58 Ga (Poiré, 1987; Gómez Peral *et al.*, 2007; Gómez Peral, 2008), and ~0.56-0.41 Ga (Iñiguez and Zalba, 1974; Gómez Peral *et al.*, 2005; Arrouy *et al.*, 2016a and b), respectively. On the top of the above-mentioned succession, the Balcarce Formation occurs. It consists of a siliciclastic sequence of Lower Cambrian to Silurian age (Teruggi *et al.*, 1958; Poiré *et al.*, 2003; Seilacher *et al.*, 2002; Zimmermann and Spalletti, 2009; Cingolani, 2010). Both successions (calcareous-siliciclastic and siliciclastic) correspond to a marine platform environment (Poiré, 1987; Poiré *et al.*, 2003).

2.3 Metamorphic and tectonic settings

The tectono-magmatic evolution of the basement occurred during the Transamazonian Cycle or its equivalent in Argentina, the Tandileano Cycle (Teruggi and Kimurray, 1975; Dalla Salda et al., 1988; Ramos, 1999). It is linked to the accretion of the Tandilia terrane, on the southern margin of the BA-PA terrane (Teruggi *et al.*, 1988; Dalla Salda et al., 1988), that generated deformation, metamorphism (amphibolite to granulite facies), and anatexis of cortical rocks (Dalla Salda and Franzese, 1989; Cingolani and Dalla Salda, 2000; Cingolani, 2010). Recently, Chernicoff et al. (2014) interpreted the evolution history previous to the 2.00 Ga collision through geophysical studies. These authors point out that a rift system initiated in the Neoarchaean-Siderian would have separated the Tandillia terrane of the RPC, generating an ocean basin (Fig. 3a). Afterwards, between 2.37 and 2.17 Ga (early Rhyasian), the beginning of a subduction towards the north of Tandilia, and another intraoceanic subduction, determined the formation of a continental magmatic arc and an island arc (El Cortijo, Teruggi et al., 1988; Cingolani, 2010), respectively (Fig. 3b). This episode ended at 2.0 Ga (late Rhyasian), with the collision of the Tandilia terrane with the southern margin of the BA-PA terrane (Fig. 3c) (Cingolani, 2010; Chernicoff et al., 2014). Then, a tensional stage occurred during the Neoproterozoic to Lower Cambrian period when an important faulting affected the Palaeoproterozoic basement and, partly, the sedimentary cover (Cingolani, 2010).

2.4 San Miguel area

The San Miguel (SM) area is located 60 km southwest of Tandil city (Fig. 2c). It comprises basement outcrops, surrounded by an extended modern sedimentary cover (Fig. 4). The most widespread rocks are biotite gneiss, migmatites (Rb-Sr age of 2.12 ± 0.01 Ga; Halpern *et al.*, 1970), calcitic marble, and a wollastonite-vesuvianite-grossular-diopside skarn (Lajoinie, 2015) (Fig. 4). Also, igneous rocks crop out in the area as tonalite and granite bodies, and an andesite-basaltic dike (Lajoinie *et al.*, 2014a).

The biotite gneiss displays a heterogeneous colouration with pale and dark tonalities of grey, medium-coarse grain size and equigranular texture. It shows a planar anisotropic fabric of compositional banding type (S_1), given by the alternation of pale and dark bands orientated in a N35°E direction and 60°SW inclination (Fig. 5a). The pale bands are whitish, approximately 2-3 cm wide and are composed of quartz, K-feldspar, and plagioclase, while the dark ones are black or grey, approximately 4 cm wide, and contain abundant biotite besides the above-mentioned minerals (Fig. 5b).

The migmatites generally exhibit stromatic structures produced by the alternation of leucosome bands (parallel to S_1 foliation) with mesosomes that corresponds to the biotite gneiss (Fig. 5c). The leucosomes are composed of quartz, K-feldspar, and plagioclase crystals of coarse grain size; also, garnets are observed in some of the more voluminous leucosomes (Fig. 5d). Commonly, these bodies exhibit a whitish to greyish colouration, with medium to coarse grain size and porphyroblastic texture in response to the presence of reddish-brownish garnet porphyroblasts up to 1 cm in diameter (Fig. 5d).

The marble (Fig. 6a) has whitish to grey colouration and is mostly composed of calcite, and a lesser extent by a green-coloured clinopyroxene. It is in contact with the gneiss and is intruded by migmatite leucosomes or leucogranites (Fig. 6a and b). Leucogranites

have a "sill" morphology that repeatedly cuts the marble, developing numerous marblelayered bodies of 1- to 2-m-wide (Fig. 6b and c). A calc-silicate paragenesis, consisting of wollastonite+vesuvianite+grossular+diopside, developed in each contact marbleleucogranite, forming the San Miguel skarn (Fig. 6d; Lajoinie *et al.*, 2013).

3. Analytical methods and sample locations

A total of 12 representative samples of the SM marble were selected for petrographic studies and eight for geochemical characterisation. The sampling pattern followed a cross-section in the marble that went from the furthest to the nearest sectors, to the contact with leucogranite bodies (Fig. 7a and b). The composition of the clinopyroxene crystals was determined using an electron microprobe Cameca, Camebax SX 100 model, in the microanalysis laboratory of the Servicio de Ciencia y Técnica de la Universidad de Oviedo, Spain. Operating conditions were 20 kv accelerating voltage, beam current of 20 nA, and beam size of 1 to 2 μ m. Natural standards certificated by MAC (Micro Analysis Consultants Ltd. United Kingdom) were used for calculating element concentrations: albite for Na and Si, Al₂O₃ for Al, MgO for Mg, andradite for Ca, orthoclase for K, synthetic Mn and Ti oxides for Mn and Ti, magnetite for Fe, NiO for Ni and chromite for Cr.

Geochemical analyses of the samples, previously crushed and pulverised to 200 mesh in a ring mill, were carried out at the Laboratorio ALS Patagonia S.A. (Argentina). Major elements (SiO₂, TiO₂, Al₂O₃, Fe₂O₃, MnO, MgO, CaO, Na₂O, K₂O, and P₂O₅) were determined by X-ray fluorescence spectrometry following the ME-XRF12 routine, where the samples were calcined and dissolved with 50% Li₂B₄O₇-50% LiBO₂, and

finally melted at 1050-1100°C until forming a pearl. The standards used were SARM-39 and SARM-45. Elements such as Ba, Cr, Rb, Cs, Ga, Hf, Nb, Y, Sr, Sn, Ta, Th, U, V, W, Zr, and REE were determined following the ME-MS81 routine, which consists of mixing 0.20 g of the sample with 0.90 g of LiBO₂ and melting them at 1000°C. The solution obtained was cooled and dissolved in 100 ml of 4% HNO₃-2% HCl and then analysed using inductively-coupled plasma mass spectrometry (ICP-MS).

Isotopic determinations of δ^{13} C and δ^{18} O were conducted at the Servicio de Isótopos Estables de la Universidad de Salamanca (Spain) using a SIRA-II gas source mass spectrometer, manufactured by VG-Isotech and equipped with *cold finger* to analyse small samples and a multiple system of sample admission. The mineral samples (calcite and diopside) were separated and concentrated by hand-picking under a binocular microscope from other fractions of the same samples used for chemical analyses. For the gas extraction (CO₂) a reaction with 103% H₃PO₄ at 25°C for 12 hours was used. Conventional methodologies according to Clayton and Mayeda (1963) were used for the treatment of the diopside crystals. The isotopic relationships are reported in delta notation (δ ‰) relative to the V-PDB (*Pee Dee Belemnite*) standard for the δ^{13} C, and to the V-SMOW (*Standard Mean Ocean Water*) for the δ^{18} O, with an analytical error of \pm 0,2‰.

Sm-Nd isotopic dating on garnet-whole rock pairs of two skarn samples was performed at the Laboratório de Geocronologia, Universidade de Brasília (Brazil), according to the methodology proposed by Gioia and Pimentel (2000).

The samples were pulverised and their Sm and Nd contents were first determined using X-ray fluorescence. The extraction of the elements (Sm and Nd) was done using conventional cation exchange techniques, and then loaded onto Re evaporation

filaments of double filament assemblies. The isotopic measurements were performed on a multicollector Finnigan MAT 262 mass spectrometer in static mode. The uncertainties for the ¹⁴⁷Sm/¹⁴⁴Nd and ¹⁴³Nd/¹⁴⁴Nd ratios are better than \pm 0.2% (2 σ) and \pm 0.005% (2 σ), respectively. ¹⁴³Nd/¹⁴⁴Nd ratios were normalised to an ¹⁴⁶Nd/¹⁴⁴Nd ratio of 0.7219 and the decay constant used was λ^{147} Sm = 6.54 × 10⁻¹² a⁻¹.

4. Petrography of marble

The marble has an inequigranular coarse-grained texture ranging from 2 mm to \sim 30 mm grain size. Its mineralogy is mostly composed of calcite, with small amounts of clinopyroxene of a diopside type (di_{62.32-79.60} hd_{20.23-37.43} jo_{<0.25}; Table 1) and scarce quartz (Fig. 8a).

The calcite crystals are well developed; they have intracrystalline deformation as evidence of undulose extinction, and tapered and deformed lamellae twins (Fig. 8b). Also, some calcite crystals show mechanical twinning of crossing and stopping types (Chen *et al.*, 2011; Fig. 8c). Diopside is dark green, with a medium crystal length of 3-4 mm and has a slightly variable abundance between 10 and 15%. Some crystals are replaced by chlorite and amphibole, of tremolite type (Fig. 8d). Scarce and isolated quartz crystals of 0.5 mm average size are present, showing undulose extinction (intracrystalline deformation) and subgrain presence (recovery) (Fig. 8b). Apatite occurs as accessory minerals.

5. Geochemical characterisation of marble

5.1 Major and trace elements

Bulk rock chemical analyses for all samples are presented in Table 2. The studied marble exhibits high contents of CaO (51.60 to 54.80 wt.%), in agreement with the great amount of calcite identified. Low SiO₂ (1.62 to 8.65 wt.%), MgO (0.53 to 1.13 wt.%), and Fe₂O₃ (0.13 to 1.63wt.%) contents are attributed to the presence of diopside and quartz crystals. In addition, low Al₂O₃ (< 1.22 wt.%), Ti₂O (<0.04 wt.%), Hf (< 0.30 ppm), Th (< 0.59 ppm), and Zr (< 14 ppm) contents residing in the terrigenous silicate fraction were found. Samples are also depleted in trace elements, except for Ba and Sr, which show a wide variation between 10.4 and 1080.0 ppm, and 115.0 and 303.0 ppm, respectively. Finally, a slightly negative correlation exists for CaO vs. Sr (Fig. 9a) and positive for Σ REE vs. Al₂O₃ (Fig. 9b) and Ti₂O vs. Zr (Fig. 9c).

5.2 REE and Y

Rare earth elements and Y data are shown in Table 3. These contents were normalised to Post-Archaean Average Australian Shale (PAAS; Taylor and MacLennan, 1985) and plotted in Fig. 10. These samples have very low \sum REE contents (7.43-53.34 ppm), slight enrichment of HREE (Pr/Yb_{PAAS} 0.55-1.00), and MREE (Sm/Yb_{PAAS}) contents of 0.67 to 1.64. Only a-SMC and SMC-1 samples show HREE contents of 1.94 and 1.05 ppm. The studied samples also have positive Eu/Eu* anomalies between 1.53 and 2.97 (Eu/Eu* according to Bau and Dulski, 1996) and truly negative Ce anomalies on the basis of Ce/Ce* <1 and Pr/Pr* >1 in all cases (following Bau and Dulski's, 1996 method). Finally, the La/La* (1.27-2.00, after Bolhar *et al.*, 2004) and Y (Y/Ho 47.06-70.84) anomalies are positive.

5.3 $\delta^{13}C$ and $\delta^{18}O$ isotopes

All δ^{18} O values obtained from marble calcite crystals are positive and range between +17.45 and +13.84‰ (V-SMOW), whereas the δ^{13} C contents are also remarkably positive and vary between +5.89 and +4.26‰ (V-PDB), as indicated in Table 4. This table also includes data obtained from Lajoinie *et al.* (2014b). The values plotted in a δ^{13} C vs. δ^{18} O diagram (Fig. 11) show the fluid evolution in isotopic equilibrium with calcite. A correlation between isotopic values and its position with respect to the marble-leucogranite contact is highlighted in the diagram. The correlation shows that the most positive values of δ^{13} C and δ^{18} O correspond to the samples located furthest from the above-mentioned contact.

 δ^{18} O (V-SMOW) data from marble diopside is shown in Table 4. The diopside δ^{18} O values (+14.90, +14.70, and +14.50‰) together with those obtained for calcite from the same samples (+17.45, +17.12, and +16.80‰ V-SMOW) allowed the isotopic equilibrium temperature to be determined using the calcite-diopside geothermometer proposed by Zheng (1993). These results indicate temperatures of 623°C to 663°C (Table 4).

5.4 Sm-Nd

Two garnet-whole rock pairs of skarn samples were analysed using the Sm-Nd method (Table 5); however, reliable ages could not be obtained. The Nd model ages related to the depleted mantle (T_{DM}) of the studied skarn are 2.47 Ga and 3.05 Ga. The model ages of sedimentary rocks reflect average crustal residence ages of their source region (Goldstein *et al.*, 1997). In this sense, the Nd model ages of skarn samples from the San Miguel area are similar to those obtained by Pankhurst *et al.* (2003) for other basement

units of the Tandilia System in the Olavarría area ($T_{DM} = 2.46-2.52$ Ga). Our results provide reliable evidence of participation of a Siderian or older basement in the formation of the studied rocks, since the real value characterising the moment of separation of material from the evolutionary line of the depleted mantle could be older (*e.g.* Arndt and Goldstein, 1987).

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6. Discussion

Palaeoproterozoic carbonates deposited during the "Lomagundi-Jatuli event" (2.22 to 2.06 Ga; Karhu and Holland, 1996) have been recognized in several formations belonging to different cratonic units around the world, due to the presence of an important positive anomaly in the $\delta^{13}C$ (\geq +5‰) values.

In the San Miguel area, one of the two unique carbonate rocks which form part of the Palaeoproterozoic crystalline basement of the Tandilia System, Río de la Plata Craton, crops out. The marble has a calcite+diopside+quartz \pm apatite mineral association and is in contact with the biotite gneiss and intruded by leucogranites (2.12 \pm 0.01 Ga Rb-Sr age; Halpern *et al.*, 1970). Considering the presence of a high-grade metamorphic rock assemblages and the diopside isograde temperature obtained (~650°C, for 8 Kbars P, Bucher and Frey, 2002), the regional metamorphic processes must have reached an amphibolite facies grade. These metamorphic processes also generated calcite crystal deformations and compositional banding (S₁) formation in the gneiss. According to the regional tectonic context of the Tandilia System, this episode corresponds to the Transamazonian Orogeny (2.20-1.80 Ga; Dalla Salda *et al.*, 2005). This assertion is supported by the Rb-Sr age of the leucogranites (consistent with the ages obtained for

other basement gneisses and migmatites), which could be considered the mínimum depositional age for the marble. Furthermore, the provenance of the marble may have a 2.47 to 3.05 Ga component or source according to the Nd model ages. In addition, Rapela *et al.* (2007), Gaucher *et al.* (2008), and Cingolani (2010) performed U-Pb ages obtained from detrital zircons near the study area. Detrital zircons were sampled from the base of the Sierras Bayas Group, more precisely from the "Lower quartzites" of the Villa Mónica Formation, which overlays the metamorphic basement rocks (Poiré, 1993). These determinations show a unimodal Palaeoproterozoic age indicating that this formation was mostly derived from the underlying basement, which has a restricted age range (Cingolani, 2010).

Although the metamorphism has generated some modifications in the chemical patterns of the studied marble, the chemical analyses show some particularities that correspond to the primitive characteristics of the marble protolith. The abundance of CaO and the low contents of the other majority oxides indicate that the marble is almost calcitic. Also, low contents of terrigenous silicate fraction elements (Al₂O₃, Ti₂O, Hf, Th, and Zr) indicate that the REE +Y values in the marble reflect marine conditions (Alexander *et al.*, 2008; Bolhar *et al.*, 2004; Bau and Dulski, 1996). Positive Eu anomalies are typical in Archaean-Palaeoproterozoic precipitates, and manifest the high hydrothermal fluid activity in the oceans for that period (Alexander *et al.*, 2008). Moreover, the presence of positive Y anomalies is also characteristic of seawater precipitates (Nosaki *et al.*, 1997), while negative Ce anomalies indicate an oxidative precipitation environment for the carbonate protolith. In these conditions, Ce⁺³ is oxidised to Ce⁺⁴ (less soluble), and is consequently removed from the solution by suspended particles (Bolhar *et al.*, 2004). In this sense, the Archaean-Palaeoproterozoic transition was

indicated as a period when free atmospheric di-oxygen became relatively abundant (Fig. 12a; Cloud, 1968; Martin *et al.*, 2013) based on the appearance of oxidised "red" beds in sedimentary successions, and the disappearance of detrital uraninite, siderite, and pyrite (Fig. 12b; Cloud, 1972; Martin *et al.*, 2013). This event, called the Great Oxidation (Holland, 2002), occurred between 2.50 and 2.32 Ga and was followed by the Lomagundi-Jatuli event (2.20-2.06 Ga) (Fig. 12).

 δ^{13} C stable isotope data obtained from calcite crystals revealed an important positive anomaly (Fig. 12c), confirming the marine origin of the marble suggested by Lajoinie et al. (2013, 2014b). According to these results, a decrease in the δ^{13} C (from +5.89 to +4.26‰) and in the δ^{18} O values (from +17.85 to +13.84‰) is observed towards the contact between the marble and leucogranite intrusions. Although many of the LJE carbonates remain unaltered, others show evidence of diagenetic and even metamorphic processes. According to several studies (Baker and Fallick, 1989; Melezhik et al., 2001; Melezhik and Fallick, 2010), the original δ^{13} C and δ^{18} O values can be modified or remain almost unaltered even under amphibolite facies conditions, depending on metamorphic conditions and on the carbonate rock compositions. In the study area, the presence of a wollastonite-vesuvianite-grossular-diopside skarn (Lajoinie et al., 2013) suggests that the calc-silicate mineral formation, produced by devolatilization reactions (Valley, 1986), may be the cause of the decreasing patterns registered in the δ^{13} C and δ^{18} O values. Normally, when calc-silicates form, CO₂ released from the carbonate is enriched in ¹³C and ¹⁸O (Shieh and Taylor, 1969). However, the δ^{13} C contents are not significantly modified given that it is more strongly influenced by carbonate rocks than by CO₂ content in leucogranite-derived fluids. In metasomatic processes, the oxygen isotope values are the most sensitive to modification, but there is up to 2 to 4%

permissible δ^{18} O depletion, called the "calc-silicate limit" (Valley, 1986). Thus the ~ +18‰ δ^{18} O values originally registered could have been around 20-22‰.

 δ^{18} O values obtained from marble diopside crystals allowed a temperature range of 663°C-623°C to be determined, using the geothermometric calcite-diopside pair, which is in accordance with the 716°C temperature obtained by Lajoinie *et al.* (2014a), and with the above-mentioned high degree of amphibolite facies, estimated for the SM outcropping rocks.

7. Conclusions

1. Positive Eu and Y anomalies, together with a slight enrichment of MREE and HREE (with respect to LREE), indicate that the SM marble protolith would have been formed by chemical precipitation from mixed hydrothermal and marine fluids during the Palaeoproterozoic era.

2. The Ce negative anomaly indicates oxidising conditions for precipitation environments, which is consistent with the relatively more-abundant free atmospheric oxygen for the period.

3. δ^{13} C and δ^{18} O values show decreasing patterns provoked by metamorphic and metasomatic processes. However, $\geq +5\%$ δ^{13} C values and closer to 18‰ δ^{18} O values together with the Palaeoproterozoic estimated age indicate that the SM marble was derived from a marine carbonate deposited during the "Lomagundi-Jatuli event".

4. The presence of Lomagundi-Jatuli marble in the Tandilia System basement constitutes a new correlation element between the Rio de la Plata Craton and other

Archaean-Palaeoproterozoic Cratons, and even among other carbonate rocks in the same Craton.

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Figure 1: Worldwide distribution of the Lomagundi-Jatuli event carbonates (modified from Melezhik *et al.*, 1999; Tang *et al.*, 2011) and San Miguel marble location (Tandilia System, Río de la Plata Craton).

Figure 2: a) Location, extension, and shape of the Rio de la Plata Craton proposed by different authors as follows: (A) Santos *et al.* (2017), (B) Bossi and Cingolani (2009), (C) Rapela *et al.* (2007), (D) Fuck *et al.* (2008), (E) Oyhantçabal *et al.* (2011) and Rapela *et al.* (2011), (F) Ramos *et al.* (2010). P-terrane: Patagonian terrane b) Schematic geological map of the RPC outcrops in Brazil Paraguay, Argentina, and Uruguay, along with the Neoproterozoic terranes, belts, and major shear zones (modified from Santos *et al.*, 2017). SYSZ: Sarandi del Yí Shear Zone, SBSZ: Sierra Ballena Shear Zone (Oyhantçabal *et al.*, 2010a), R: Rivera, T: Tacuarembó, LA: Luís Alves, Ca: Camboriú, Cu: Curitiba, Sg: Sao Gabriel arc. BA-PA correspond to Buenos Aires-Piedra Alta terrane (Chernicoff *et al.*, 2014). c) Geological map of the Tandilia System (Dalla Salda *et al.*, 2005), the study area is highlighted (more detail in Fig. 4).

Figure 3: Schematic diagrams of the interpreted tectonic evolution for the Tandilia System (from Cingolani, 2010; Chernicoff *et al.*, 2014): a) rift system initiation; b) beginning of subduction (after Teruggi *et al.*, 1988; Ramos, 1999); and c) main collisions, crust thickening, transcurrent faults, anatexis.

Figure 4: Geological map of the study area. Modified from Lajoinie et al. (2014b).

Figure 5: a) Biotite gneiss outcrops. b) Detail of the gneiss banded structure, LG = leucocratic gneiss, MG = melanocratic gneiss. c) Migmatite with stromatic structure outcrop. d) Detail of garnet-leucosome in contact with the gneiss.

Figure 6: a) Marble outcrop cut by leucosome intrusions. b) Schematisation of the leucosome-marble outcropping relationships. c) Detail of the marble lenses and the leucosome intrusions. d) Wollastonite-vesuvianite-grossular-diopside skarn. Mineral abbreviations according to Kretz (1983).

Figure 7: Sample locations. a) Cross-section in the marble from the furthest to the nearest sectors to the contact with leucogranite bodies. b) Cross-section between two leucogranite intrusions.

Figure 8: SM marble microscopy aspect. a) Prismatic and basal sections of diopside crystal surrounded by calcite. b) Intracrystalline deformation evidence of tapered and deformed lamellae twins in calcite, and undulose extinction in quartz. c) Two types of mechanical twining: crossing (CT) and stopping (ST). d) Diopside crystal replacements by tremolite and chlorite. Also, more evidence of deformation as undulose extinction (UE) and tapered twins in calcite are shown. Mineral abbreviations according to Kretz (1983).

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Figure 9: Geochemical binary diagram (marble samples): a) CaO vs. Sr; b) \sum REE vs. Al₂O₃; and c) Ti₂O vs. Zr. The first diagram shows a slightly negative relation, while the other two exhibit positive correlations.

Figure 10: Post-Archaean Average Australian Shale (PAAS; Taylor and McLennan, 1985) normalised rare earth elements and yttrium data from SM marble.

Figure 11: δ^{18} O *vs.* δ^{13} C values of the calcite crystals from the SM marble. The boxes correspond to typical values of: Lomagundi-Jatuli carbonates (LJC, Martin *et al.*, 2013), normal marine carbonates (NMC), and igneous calcites (IC), according to Bowman (1988).

Figure 12: Important events during the Archaean-Palaeoproterozoic transition (Martin *et al.*, 2013). a) Rise in oxygen levels between 2.45 and 2.20 Ga. %P.A.L.: present atmospheric levels. Reducing (dark boxes) and oxidising (grey boxes) atmospheres are plotted (Holland, 1994; Bekker and Holland, 2012). b) Distribution of Fe deposits (Isley and Abbott, 1999). Δ^{33} S values according to Farquhar and Wing (2003) and Reinhard *et al.* (2013). c) δ^{13} C from marine carbonates (Shields and Veizer, 2002). Also, SM marble is plotted.

























		clir	nopyroxene d	composition				
	P1	P2	P3	Р4	Р5	P6	P7	P8
wt.%								
SiO ₂	52.82	53.87	53.12	53.16	53.25	52.05	52.63	53.08
TiO ₂	0.00	0.02	0.04	0.00	0.04	0.07	0.00	0.02
AI_2O_3	0.63	0.53	0.63	0.7	0.66	0.31	0.63	0.17
Cr_2O_3	0.01	0.04	0.03	0.02	0	0	0	0.03
FeO*	6.99	5.39	5.48	6.62	5.37	10.93	8.65	8.79
$Fe_2O_3^*$	1.71	1.15	1.68	1.65	1.43	0.76	1.66	0.93
MnO	0.01	0.05	0.04	0.06	0.02	0.08	0.00	0.04
MgO	12.85	14.25	13.84	13.16	14.02	10.89	11.63	12.39
NiO	0.03	0	0.07	0	0	0.06	0	0.01
CaO	25.34	25.65	25.74	25.7	25.59	24.78	25.76	25.61
Na₂O	0.21	0.19	0.16	0.17	0.18	0.09	0.18	0.08
K ₂ O	0.03	0.04	0.01	0	0	0.01	0.01	0.03
P_2O_5	0.16	0.17	0.14	0.11	0.18	0.17	0.14	0.21
Σ	100.79	101.35	100.97	101.35	100.74	100.2	101.28	101.38
cations			(6 oxygens				
Si	1.96	1.97	1.96	1.96	1.97	1.98	1.96	1.97
Р	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al	0.03	0.02	0.03	0.03	0.03	0.01	0.03	0.01
Fe ⁺³	0.01	0	0.01	0.01	0	0.01	0.01	0.02
Σ	2.00	2.00	2.00	2.00	2.00	2.00	2.00	2.00

Table 1: Representative electron microprobe analyses of clinopyroxene crystals from SM marble.

Ті	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Cr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Fe ⁺³	0.05	0.03	0.04	0.04	0.04	0.02	0.05	0.02
Fe ⁺²	0.21	0.16	0.16	0.2	0.16	0.34	0.27	0.26
Mn	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Mg	0.71	0.78	0.76	0.72	0.77	0.62	0.65	0.69
Ni	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Na	0.02	0.01	0.01	0.01	0.01	0.01	0.01	0.01
К	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Са	1.01	1.01	1.02	1.02	1.01	1.01	1.03	1.02
Σ	2.00	2.00	2.00	2.00	2.00	2.00	2.00	2.00
∑ _{Total}	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00
			Molecul	ar %				
Di	72.76	79.6	77.77	74.1	78.85	62.32	67.08	69.46
Hd	27.21	20.23	22.12	25.71	21.08	37.43	32.92	30.4
ol	0.03	0.17	0.12	0.19	0.08	0.25	0	0.14
Σ	100	100	100	100	100	100	100	100

* Calculated following Droop (1987).

			San	vinguer m	laidie, Ta	inuma sy	stem		
Sample #	SMC-1	SMC-2	SMC-3	SMC-4	SMC-5	SMC-6	SMC-7	a-SMC	b-SMC
wt.%									
SiO ₂	2.18	2.54	4.57	3.28	4.16	6.59	4.08	8.65	1.62
Al_2O_3	0.32	0.22	0.46	0.6	0.69	0.3	0.23	1.22	0.0
Fe ₂ O _{3total}	0.33	0.46	0.66	0.32	0.37	0.3	0.66	1.63	0.1.
CaO	54.8	56.1	52.9	56.2	53.3	55	54.4	51.6	54.:
MgO	0.53	0.57	1.1	0.57	0.62	0.95	1.13	1.12	0.6
Na ₂ O	0.09	0.05	0.05	0.18	0.1	0.09	0.1	0.14	0.04
K ₂ O	0.04	0.05	0.01	0.05	0.02	0.03	0.02	0.03	0.0
TiO ₂	0.01	0.01	0.01	0.02	0.03	< 0.01	0.01	0.04	< 0.0
MnO	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.03	0.0
P_2O_5	0.1	0.09	0.07	0.1	0.09	0.11	0.06	0.09	0.0′
LOI	40.7	40.5	40.4	40.5	40.5	35.9	39.7	34.7	42.9
Σ	99.12	100.08	100.27	101.84	100.03	99.29	100.41	99.28	100.0
Ppm									
Ba	25.6	28.1	85.4	40.4	1080	22.1	21.1	37.1	10.4
Cr	10	<l.d< td=""><td><l.d< td=""><td>10</td><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td>10</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<></td></l.d<></td></l.d<>	<l.d< td=""><td>10</td><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td>10</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<></td></l.d<>	10	<l.d< td=""><td><l.d< td=""><td><l.d< td=""><td>10</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<>	<l.d< td=""><td><l.d< td=""><td>10</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<>	<l.d< td=""><td>10</td><td><l.i< td=""></l.i<></td></l.d<>	10	<l.i< td=""></l.i<>
Cs	0.07	0.08	0.02	0.03	0.07	0.02	0.02	0.09	0.0
Ga	1.4	1.3	0.6	2	0.8	1.3	1.7	2.8	<l.i< td=""></l.i<>
Hf	< 0.2	<l.d< td=""><td><l.d< td=""><td>0.2</td><td>0.3</td><td>< 0.2</td><td>0.2</td><td>0.5</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<>	<l.d< td=""><td>0.2</td><td>0.3</td><td>< 0.2</td><td>0.2</td><td>0.5</td><td><l.i< td=""></l.i<></td></l.d<>	0.2	0.3	< 0.2	0.2	0.5	<l.i< td=""></l.i<>
Nb	0.4	0.6	0.4	0.8	0.4	0.4	0.3	1.1	0.
Rb	0.6	0.3	<l.d< td=""><td>0.6</td><td>0.4</td><td>0.5</td><td>0.3</td><td>0.3</td><td><l.i< td=""></l.i<></td></l.d<>	0.6	0.4	0.5	0.3	0.3	<l.i< td=""></l.i<>
Sn	<l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td>2</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<></td></l.d<></td></l.d<></td></l.d<></td></l.d<>	<l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td>2</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<></td></l.d<></td></l.d<></td></l.d<>	<l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td>2</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<></td></l.d<></td></l.d<>	<l.d< td=""><td><l.d< td=""><td><l.d< td=""><td><l.d< td=""><td>2</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<></td></l.d<>	<l.d< td=""><td><l.d< td=""><td><l.d< td=""><td>2</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<>	<l.d< td=""><td><l.d< td=""><td>2</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<>	<l.d< td=""><td>2</td><td><l.i< td=""></l.i<></td></l.d<>	2	<l.i< td=""></l.i<>
Sr	155.5	133.2	191.5	162	166.5	169	196	303	11:
Та	0.1	0.1	<l.d< td=""><td>0.1</td><td>0.1</td><td>0.4</td><td>0.2</td><td>0.2</td><td><l.i< td=""></l.i<></td></l.d<>	0.1	0.1	0.4	0.2	0.2	<l.i< td=""></l.i<>
Th	0.41	0.32	0.33	1.31	0.59	0.32	0.26	0.67	0.12
U	0.22	0.15	0.19	0.11	0.15	0.16	0.14	0.65	0.0
V	6	5	<l.d< td=""><td><l.d< td=""><td><l.d< td=""><td>5</td><td><l.d< td=""><td>6</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<></td></l.d<>	<l.d< td=""><td><l.d< td=""><td>5</td><td><l.d< td=""><td>6</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<></td></l.d<>	<l.d< td=""><td>5</td><td><l.d< td=""><td>6</td><td><l.i< td=""></l.i<></td></l.d<></td></l.d<>	5	<l.d< td=""><td>6</td><td><l.i< td=""></l.i<></td></l.d<>	6	<l.i< td=""></l.i<>
W	33	28	<l.d< td=""><td>19</td><td>0.1</td><td>87</td><td>33</td><td>35</td><td><l.i< td=""></l.i<></td></l.d<>	19	0.1	87	33	35	<l.i< td=""></l.i<>
7	1	2	2	9	14	3	3	23	<l.i< td=""></l.i<>

Table 2: Major, minor and trace element compositions for all the marble study samples.

Locality			San N	liguel ma	arble, T	andilla S	ystem		
Sample #	SMC-1	SMC-2	SMC-3	SMC-4	SM-5	SMC-6	SMC-7	a-SMC	b-SMC
La	5.73	5.03	3.82	4.97	5.35	6.11	3.82	14.13	1.15
Ce	4.78	4.67	3.98	4.78	5.57	4.78	3.18	15.12	0.80
Pr	0.88	0.91	0.62	0.88	0.97	0.88	0.62	1.85	0.18
Nd	3.73	2.88	2.71	3.39	3.73	3.05	2.71	7.12	1.02
Sm	0.78	0.67	0.39	0.67	0.44	0.44	0.56	1.00	0.17
Eu	0.32	0.33	0.17	0.25	0.28	0.14	0.21	0.38	0.05
Gd	0.85	0.87	0.58	0.58	0.94	0.62	0.71	1.20	0.18
Tb	0.10	0.09	0.08	0.11	0.09	0.06	0.08	0.14	0.03
Dy	0.84	0.77	0.47	0.75	0.75	0.51	0.61	0.98	0.28
Y	10.26	10.33	6.21	9.72	9.72	7.02	8.10	10.26	3.24
Но	0.18	0.16	0.11	0.17	0.18	0.10	0.16	0.22	0.05
Er	0.43	0.44	0.29	0.51	0.48	0.37	0.37	0.48	0.14
Tm	0.08	0.07	0.07	0.06	0.10	0.07	0.07	0.10	0.03
Yb	0.28	0.33	0.20	0.31	0.34	0.31	0.37	0.31	0.11
Lu	0.05	0.03	0.02	0.03	0.03	0.02	0.05	0.03	0.01
∑REE	29.29	27.58	19.71	27.17	28.98	24.50	21.61	53.34	7.43
Ү/Но	57.52	64.56	56.97	57.70	54.49	70.84	51.08	47.06	65.39
LREE(Pr/Yb) _{PAAS}	1.00	0.88	1.00	0.91	0.92	0.91	0.54	1.91	0.50
MREE(Sm/Yb) _{PAAS}	1.40	1.03	1.00	1.09	0.67	0.73	0.77	1.64	0.75
Eu/Eu* _{sN}	2.40	2.97	2.13	2.09	2.74	1.53	2.11	1.79	1.67
Ce/Ce* _{sN} Ce _{sN} /(0.5La _{sN} +0.5Pr _{sN})	0.48	0.50	0.59	0.52	0.56	0.46	0.47	0.66	0.40
Ce/Ce* _{sN} Ce _{sN} /(2Pr _{sN} +Nd _{sN})	0.19	0.20	0.23	0.20	0.21	0.21	0.18	0.30	0.17

Table 3: Rare earth element and yttrium data (ppm) for all the marble study samples.

Pr/Pr* _{sN} Pr _{sN} /(0.5Ce _{sN} +0.5Nd _{sN})	1.18	1.44	1.08	1.25	1.22	1.33	1.17	1.05	1.33
La/La* _{sN} La _{sN} /(3Pr _{sN} - 2Nd _{sN})	1.88	0.95	2.00	1.30	1.27	1.33	2.00	1.76	1.50
SN: Normalized to shale.									Ò
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_	Sample	$\delta^{18}O_{cal}$ ‰	$\delta^{13}C_{cal}$ %	$\delta^{18}O_{clpx}$ %	T °C	
	SMC-1	+17.45	+5.89	+14.50	623	
	SMC-2	+17.12	+5.42	+14.30	643	
	SMC-3	+16.80	+5.39	+14.10	663	
	SMC-4	+16.80	+5.06	n.d.		
	SMC-5	+13.96	+5.05	n.d.		
	SMC-6	+13.91	+4.95	n.d.		
	SMC-7	+13.84	+4.26	n.d.	C	
	a-SMC	+17.15	+5.42	n.d.		
	b-SMC	+16.81	+4.94	n.d.	. 67	
	c-SMC-	+17.05	+5.40	n.d.		
					· · · · · · · · · · · · · · · · · · ·	

Table 4: Isotope composition of SM marble. δ^{13} C and δ^{18} O values from calcite relative to V-PDB (Vienna Pee Dee Belemnite) and to V-SMOW (Standard Mean Ocean Water), respectly. Also, δ^{18} O (V-SMOW) from diopside is presents.

The SMC-7 to SMC-1 samples follow a core to rim marble cross-section, and a-SMC, b-SMC and c-SMC correspond to a rim-core-rim marble lenses sampling.

Sample Sm* Nd* ¹⁴⁴ Nd ¹⁴⁴ Nd ¹⁴⁴ Nd); εNd ₍₀₎ T _{DM} (Ga Grt _{5M-22} 2.15 8.43 0.153973 0.511635 0.510501 - - WR _{5M-29} 2.03 9.03 0.136189 0.511504 ± 0.000009 -22.12 3.05 Grt _{5M-31} 0.81 5.72 0.085908 0.511031 0.481599 - - WR _{5M-31} 0.79 5.56 0.085707 0.510962 ± 0.000024 32.69 2.47	Sample Sm* Nd* ¹⁴⁴ Nd ¹⁴⁴ Nd ¹⁴⁴ Nd); εNd ₍₀₎ T _{DM} (Ga) Grt _{5M.29} 2.15 8.43 0.153973 0.511635 0.510501 - - WR _{5M.29} 2.03 9.03 0.136189 0.511504 ± 0.00009 -22.12 3.05 Grt _{5M.31} 0.81 5.72 0.085908 0.511031 0.481599 - - WR _{5M.31} 0.79 5.56 0.085707 0.510962 ± 0.000024 32.69 2.47								
Grt _{5M-29} 2.15 8.43 0.153973 0.511635 0.510501 - - WR _{5M-29} 2.03 9.03 0.136189 0.511504 ± 0.000009 -22.12 3.05 Grt _{5M-31} 0.81 5.72 0.085908 0.511031 0.481599 - - WR _{5M-31} 0.79 5.56 0.085707 0.510962 ± 0.000024 32.69 2.47 m and Nd contents are given in ppm 	Grt _{sM-29} 2.15 8.43 0.153973 0.511635 0.510501 - - WR _{SM-29} 2.03 9.03 0.136189 0.511504 ± 0.000009 -22.12 3.05 Grt _{SM-31} 0.81 5.72 0.085908 0.511031 0.481599 - - wR _{SM-31} 0.79 5.56 0.085707 0.510962 ± 0.000024 32.69 2.47 n and Nd contents are given in ppm - - - - - -	Sample	Sm*	Nd*	¹⁴⁴ Nd	¹⁴⁴ Nd	¹⁴⁴ Nd) _i	εNd ₍₀₎	Т _{рм} (Ga)
WR _{5M-29} 2.03 9.03 0.136189 0.511504 ± 0.000009 -22.12 3.05 Grt _{5M-31} 0.81 5.72 0.085908 0.511031 0.481599 - - WR _{5M-31} 0.79 5.56 0.085707 0.510962 ± 0.000024 32.69 2.47 m and Nd contents are given in ppm	WR _{5M-29} 2.03 9.03 0.136189 0.511504 ± 0.00009 -22.12 3.05 Grt _{5M-31} 0.81 5.72 0.085908 0.511031 0.481599 - - WR _{5M-31} 0.79 5.56 0.085707 0.510962 ± 0.000024 -32.69 2.47	Grt _{SM-29}	2.15	8.43	0.153973	0.511635	0.510501	-	-
Grt _{5M-31} 0.81 5.72 0.085908 0.511031 0.481599 - - - WR _{5M-31} 0.79 5.56 0.085707 0.510962 ± 0.000024 -32.69 2.47 m and Nd contents are given in ppm	Grt _{5M-31} 0.81 5.72 0.085908 0.511031 0.481599 32.69 2.47 wR _{5M-31} 0.79 5.56 0.085707 0.510962 ± 0.000024 32.69 2.47 n and Nd contents are given in ppm	WR _{SM-29}	2.03	9.03	0.136189	0.511504	± 0.000009	-22.12	3.05
WR _{5M-31} 0.79 5.56 0.085707 0.510962 ± 0.000024 32.69 2.47 n and Nd contents are given in ppm	WR _{5M-31} 0.79 5.56 0.085707 0.510962 ± 0.000024 32.69 2.47	Grt _{sM-31}	0.81	5.72	0.085908	0.511031	0.481599	-	-
n and Nd contents are given in ppm	n and Nd contents are given in ppm	WR _{SM-31}	0.79	5.56	0.085707	0.510962	± 0.000024	-32.69	2.47
		m and Nd	contents	are give	en in ppm				
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Table 5: Results of the Sm-Nd determinations in garnet-whole rock pairs from the San Miguel skarn.

Highlights

- $\delta^{13}C > 5\%$ shows that San Miguel marble derived from a "Lomagundi-Jatuli" carbonate.
- δ¹³C and δ¹⁸O depletion was caused by Palaeoproterozoic metamorphic processes.
- Positive Y anomalies and HREE enrichment reflect marine precipitation conditions.
- Positive Eu anomalies manifest hydrothermal fluid participation in seawater.
- Ce negative anomaly indicates oxidizing precipitation conditions.