Protective Action of lons against Cadmium Toxicity to Young *Bufo arenarum* Tadpoles¹

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Several studies have shown that a large number of capable of tolerating a wide amphibian species are range external osmolarities (Bentley 1971; of Duellman and Trueb 1986). However, the mechanisms which respond to variations in the osmotic environment different in larvae and adults. They are dependent the way that are also on the osmotic i.e., through substances which is reached, pressure behave as ionic or no-ionic ones in. solution (Balinsky 1981).

eurihalinity of tadpoles was found in species of The Anura. The tolerated concentration both Urodela anđ rate was wide, from distilled water to several of mOsm. In the particular hundreds case of Bufo demonstrated young arenarum, we have that its tadpoles are able to survive when immersed in NaC1 distilled water, anđ mannitol solutions o£ considerably higher osmolarities than its natural environment (Ferrari and Salibián 1987; Ferrari et al 1988), suggesting the existence o£ adaptive physiological mechanisms.

We evaluated the impact of Cd(II) on those mechanisms in the present study. Cadmium is a contaminant widely distributed in freshwater ecosystems (Ravera 1984) known as an element that interferes in epithelial and ionic permeability processes (Hayashi et al. 1977; Hillyard 1979; Takada and Hayashi et al. 1980). We hypothesized that as а consequence o£ those properties, Cd(II) may alter the already observed ability ٥£ Bufo arenarum tadpoles to overcome important changes in the physicochemical characteristics their o£ incubation media. We evaluated the effect of a sublethal concentration of this metal on the water balance of the animals

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immersing them in different solutions. Evidence is presented that inorganic ions have shown a protective action to the toxic effects of Cd(II).

MATERIALS AND METHODS

Ovulation of <u>Bufo</u> arenarum females was induced by injection of a suspension of homologous hypophysis into the caelomic cavity (Pisanó 1956). Oocytes were fertilized in vitro with a sperm suspension made in Holtfreter 10% solution. The composition of the stock Holtfreter solution was as follows (in g/L): NaCl KCl 0.05; CaCl₂ 0.10 and NaCO₃H 0.02 3.50; (Hamburger 1969).Embryos were kept at room temperature until they reached the first larval stages (stages 26/27) (Echeverría and Fiorito de López 1981). Larvae were then transferred to artificial pond water (APW) the composition of which was (in g/L): NaCl 7.60; KCl 0.75; CaCl₂ 8.90 and NaCO₃H 1.70 (Alvarado and Johnson 1966). Tadpoles were fed every other day with pulverized fish food containing 35% protein.

All tests were conducted with animals acclimated 48 hr before the experiments began in a Ghilon TC 120 incubation chamber at constant temperature ($20 \pm 1.0^{\circ}$ C) and photoperiod (12/12 hr) and remained under the same conditions throughout the experiments.For the LC50 determination the rate of one animal per 10 mL of solution was kept constant (APHA 1975); in water balance experiments the rate was one tadpole per 4 mL.

In order to establish the LC50, we assayed the effect of the following concentrations of Cd(II): 0.5, 1.0, 2.0 and 4.0 mg/L, under semistatic conditions. These solutions were freshly prepared every day dissolving appropiate amounts of a stock CdCl₂ solution (analytical grade) in APW. Animals in APW (5 mOsm) were considered as controls. The number of animals in each glass container was 10 for each concentration run in quadruplicate (n=40). During the 96 hr test, all solutions were renewed daily, dead animals removed and their number recorded. A specimen was considered dead when its heartbeat had stopped and/or showed no response to gentle prodding with a glass it rod. During the period of exposure to different concentrations of Cd(II), the behavioral responses and somatic alterations of the tadpoles were also recorded. The LC50 values were estimated using a probit analysis program (Finney 1971; Herrera and Mavrich 1981).

For water balance experiments, we chose, from the LC50 measurements, the concentration of 1.0 mg/L of

Cd(II) as a sublethal dose. Animals of the same age were exposed in duplicate in groups of 120 tadpoles per concentration (n=240) to the following solutions containing 1.0 mg/L of Cd(II): a) deionized water (DW; 0 mOsm) pH=5.3; b) APW (5 mOsm) pH=6.7; c) APW mOsm) pH=6.8; d) sodium chloride (NaCl 141 mOsm) (30 pH=6.8; e) mannitol (MAN 141 mOsm) pH=5.4; control tests were run simultaneously, incubating tadpoles in the same solutions without Cd(II). Solutions were changed daily. Sampling was done every 24 hr during 4 days; at each time and experimental condition, 8 samples (consisting of 4 animals each) were taken. In these samples, wet weight (ww) was determined and dry weight (dw) was determined after drying at 100-105⁰C. From the data of ww and dw water content (wc, expressed as $mg H_2O/mg dw$) and humidity (H, as percentage) were calculated. Results are expressed as means + S.E.M of eight samples with larvae each. The statistical analysis was carried 4 out by means of a two-way analysis of variance (Cappelletti 1983; Snedecor and Cochran 1969)

RESULTS AND DISCUSSION

Effects of the assayed concentrations of Cd(II) on the survival of <u>Bufo arenarum</u> tadpoles are presented in Figure 1. As it can be seen, at 0.5 and 1.0 mg/L of Cd(II), mortality was less than 10%, which is an acceptable value for control animals (APHA (1975); above these concentrations toxicity showed an abrupt rise.

At the test concentrations used in this study tadpoles showed the following behavioral disorders and somatic alterations: swimming in atypical position, loss of equilibrium, increased irritability, arhythmic contractions, mild axial incurvation and epithelial peeling. The intensity of these signs was proportional to the Cd(II) concentrations.

The 96-hr LC50 for <u>Bufo</u> <u>arenarum</u> tadpoles exposed to Cd(II) was 2.08 mg/L (Table 1). This value is similar to that found by Rao and Madhyastha (1987) with tadpoles of <u>Microhyla</u> <u>ornata</u>. This level was higher than the values determined for embryos of <u>Bufo</u> <u>arenarum</u> (Pérez-Coll et al.1985,1986); from these results we conclude that the susceptibility of this anuran to cadmium is greater in the early stages of their development. The signs of toxicity observed a few hours after the beginning of the experiments provide additional evidence of the sensitivity of young tadpoles of <u>Bufo</u> <u>arenarum</u> to Cd(II). Since the tadpoles were born and bred in the laboratory in Cd(II)-free solutions, their response can be



Figure 1. Survival of <u>Bufo</u> <u>arenarum</u> tadpoles (stages 26/27) exposed to different concentrations of Cd(II). (n= 40 per concentration).

Table 1. LC50 values of Cd(II) for young <u>Bufo</u> arenarum tadpoles at different time exposures.

Time (hr)	LC50 (mg/L)	Confidence Limits	Slope	Chi Square	Correl. Coeff.
24	3.34	2.94 - 3.38	5.85	0.05	0.99
48	2.52	2.22 - 2.86	6.01	0.51	0.99
72	2.23	1.96 - 2.54	5.59	2.20	0.98
96	2.08	1.83 - 2.40	5.13	1.44	0.99

devoid of any spurious effect that could considered be attributed to previous exposure to the metal. determine the impact of Cd(II) on water In tests to balance of tadpoles, we found that mortality in always less than 10% . All control containers. was in non-ionic solutions (MAN) or tadpoles incubated media (DW) containing 1.0 mg/L of Cd(II)were ion-free dead within 12 hr of treatment. Animals in ionic with Cd(II) had mortality and NaCl) solutions (APW comparable to that of their respective controls.

experiments on water balance are shown in Results of were no statistically significant Table 2. There differences ο£ various parameters in animals the solutions of APW 5 mOsm, APW 30 mOsm control between

Table 2.	Rai dr) are	nge of n v weight enarun j	nean val t (dw), incubate	ues (8 samples water content d with (e) and	;total n (wc) and without	of animals=32 humidity (H%) (c) 1.0 mg/L	<pre>per day) of young cd(II).</pre>	of wet we larvae o:	eight (ww), E Bufo
		3 3 0	(mg) e	α α α α	(mg) e	wc (mg wate c	r/mg dw) e	o	Н % Ф
		41.51 +2	41.29	1.61	1.57 +0	21.66	19.79 +0	95,55 + 2	95.11 +0
S mosm	Г. 1	to 52.08 = 1.81	47.09 (ns)	1.84 1.84 F1 = 0.24	2.12 (ns)	29.67 FI = 0.38 (1)	28.17 ns)	96.70 F1 = 0.10	96.53 8 (ns)
		41.15	42.75	1.57	1,85	18.92	19.14	94,94	95.00
APW 30 mosm	L T	to 47.52 = 0.14	to 47.42 (ns)	to 2.28 Fl = 0.00	to 2.13 (ns)	to 26.89 Fl = 0.50 (to 23.63 ns)	$\begin{array}{c} 10 \\ 96.23 \\ F1 = 0.2 \end{array}$	to 95.86 2 (ns)
NaCl 141 mosm	- - -	32.55 to 38.66 = 0.00	34.18 to 38.05 (ns)	1.34 1.34 to 1.71 Fl = 2.44	1.25 to 1.73 (ns)	20.27 20.27 25.74 Fl = 3.41 (:	21.08 to 26.38 ns)	95.28 50 16.16 Fl = 2.2	95.05 to 96.32 2 (ns)
F1 = NS =		F of Sr not sig	nedecor gnifican	between treat t	ments				

and NaCl 141 mOsm and those from parallel experimental ones with 1.0 mg/L Cd(II). The experimental series of MAN and DW were not included in the statistical analysis because of its obvious results.

We demonstrated that young tadpoles of <u>Bufo arenarum</u> can overcome the noxious effects of the presence of 1.0 mg/L of Cd(II) when incubated in ionic solutions, whereas the same concentration of metal in ion-free or non-ionic solutions caused 100% mortality of the animals during a few hours period.

This toxicity of cadmium could be interpreted as a secondary consequence of the known inhibitory effect on epithelial ATPases. The results shown in Table 2 suggest that under our test conditions the inhibition was not associated to alterations in the water balance of the animals.

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