Latin American Journal of Pharmacy
(formerly *Acta Farmacéutica Bonaerense)*
Lat. Am. J. Pharm. **31** (4): 632-6 (2012)

Short communication
Received: May 9, 2012
Revised version: May 23, 2012
Accepted: May 25, 2012

Solubility of Losartan Potassium in Several Mono-Solvents at Different Temperatures

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***SUMMARY*. The solubilities of losartan potassium in eight solvents; 1,4-dioxane, acetonitrile, propylene glycol, N-methyl-2-pyrrolidone, 1-butanol, 1-hexanol, 1-heptanol, and 1-octanol, were measured at temperatures ranging from 293.2 to 318.2 K at atmospheric pressure using shake-flask method of Higuchi and Connors. N-methyl-2-pyrrolidone dissolved the highest amount of losartan potassium and 1,4-dioxane possessed the lowest solubilization power among the investigated solvents. The generated data was used to calculate the thermodynamic parameters of the system using the modified van't Hoff equation and the Gibbs free energy data was correlated using Abraham solvation parameters.**

KEY WORDS: Losartan potassium, Solubility, Thermodynamics.

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ISSN 0326-2383