# The cuticular hydrocarbons of the *Triatoma sordida* species subcomplex (Hemiptera: Reduviidae)

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The cuticular hydrocarbons of the Triatoma sordida subcomplex (Hemiptera: Reduviidae: Triatominae) were analysed by gas chromatography and their structures identified by mass spectrometry. They comprised mostly n-alkanes and methyl-branched alkanes with one-four methyl substitutions. n-alkanes consisted of a homologous series from C21-C33 and represented 33-45% of the hydrocarbon fraction; n-C29 was the major component. Methyl-branched alkanes showed alkyl chains from C24-C43. High molecular weight dimethyl and trimethylalkanes (from C35-C39) represented most of the methyl-branched fraction. A few tetramethylalkanes were also detected, comprising mostly even-numbered chains. Several components such as odd-numbered 3-methylalkanes, dimethylalkanes and trimethylalkanes of C37 and C39 showed patterns of variation that allowed the differentiation of the species and populations studied. Triatoma guasayana and Triatoma patagonica showed the most distinct hydrocarbon patterns within the subcomplex. The T. sordida populations from Brazil and Argentina showed significantly different hydrocarbon profiles that posed concerns regarding the homogeneity of the species. Triatoma garciabesi had a more complex hydrocarbon pattern, but it shared some similarity with T. sordida. The quantitative and qualitative variations in the cuticular hydrocarbons may help to elucidate the relationships between species and populations of this insect group.

Key words: Triatoma garciabesi - Triatoma guasayana - Triatoma patagonica - cuticle hydrocarbons - mass spectrometry

Triatomines (Hemiptera: Reduviidae: Triatominae) comprise more than 140 blood-sucking insect species; some of which are important vectors for Trypanosoma cruzi, the causative agent of Chagas disease. Triatomine species are grouped into several complexes according to their morphology, habitat and ecology (Carcavallo et al. 2000, Schofield & Galvão 2009). The cuticular hydrocarbon pattern of the three major vectors of Chagas disease, Triatoma infestans, Rhodnius prolixus and Triatoma dimidiata, as well as those of Triatoma mazzotti and Triatoma pallidipennis, two important vectors from Mexico, has been analysed by capillary gas chromatography coupled to mass spectrometry (CGC-MS) (Juárez & Brenner 1987, Juárez & Blomquist 1993, Juárez et al. 2001, Calderón-Fernández et al. 2011). Triatomine hydrocarbons consist of a complex blend of saturated straight chain alkanes containing between 18-33 carbon atoms together with methyl-branched components with one-four methyl substitutions inserted into alkyl chains ranging from 25 to more than 43 carbons. Cuticle hydrocarbons participate in several aspects of insect survival and fitness. In T. infestans, the inhibition of hydrocarbon synthesis was positively correlated with a higher susceptibility to contact insecticides (Juárez 1994). In addition,

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MPJ and GMCF are members of the CONICET Researcher Career. + Corresponding author: mjuarez@isis.unlp.edu.ar Received 2 January 2013 Accepted 29 May 2013 pyrethroid resistance was associated with a two-fold increase in the cuticular hydrocarbon content (Pedrini et al. 2009) and reduced insecticide penetration (Juárez et al. 2010). The Triatoma sordida subcomplex includes the following four species: T. sordida, which ranges from Central Brazil throughout most parts of Paraguay and Bolivia to Central Argentina, Triatoma guasayana, which is distributed throughout Central and Northern Argentina as well as most parts of Bolivia and Paraguay, Triatoma garciabesi, which can only be found in Central and Northern Argentina and Triatoma patagonica, which is exclusively distributed in Argentina, from the Patagonia region to the central part of the country (Lent & Wygodzinsky 1979, Jurberg et al. 1998). These species show a variable degree of association with humans and their houses. T. patagonica and T. guasavana are both mainly sylvatic, but they can form small colonies in the peridomestic habitat; also they were shown entering in human dwellings attracted to light or attacking humans in the field. T. sordida and T. garciabesi can form large colonies in peridomestic habitats (Diotaiuti et al. 1993, Wisnivesky-Colli et al. 1993). As a part of a larger project focused on detecting structural differentiation in the triatomine hydrocarbons, the aim of this work was to analyse the cuticle hydrocarbon pattern of the T. sordida subcomplex by CGC-MS analysis.

### MATERIALS AND METHODS

Insects and sampling sites - Adult male and female specimens of each species were analysed. *T. garciabesi* and *T. guasayana* specimens came from Santiago del Estero (Argentina) and *T. patagonica* specimens came from a colony originally from Santa Fé (Argentina). Panzera et al. (1997) reported genetic differences between Brazilian and Argentinean populations of *T. sordida*. Thus, two *T. sordida* colonies were analysed; one from Rondonópolis (Brazil) and the other from Córdoba (Argentina). Insects were  $\geq 1$  month old at the time of the analysis.

*Cuticle lipid extraction and hydrocarbon purification* - Male and female specimens were analysed separately. Samples (5 insects each) were washed with redistilled water to remove any water-soluble contaminants. Then, they were transferred to a glass tube and submerged three times in redistilled hexane (Carlo Erba Reagents, Milano, Italy) for 5 min each to extract the total lipids. The hexane volume was reduced under a nitrogen stream and the hydrocarbons were separated from the other lipid components by adsorption chromatography on a mini-column [10 mm x 5 mm internal diameter (ID)] of activated Biosil A (Bio-Rad Laboratories, Richmond, CA) and eluted with redistilled hexane (4 mL).

CGC-MS analysis - The volume of the hydrocarbon extracts was reduced under a nitrogen stream and analysed by CGC-MS using a Hewlett-Packard 6890 (Hewlett Packard, Wilmington, DE) CGC to an Agilent 5975C VL (Santa Clara, CA) mass spectrometer and interfaced with an Agilent MSD Chem Station. The injection port was operated in splitless mode at 320°C. A non-polar fused silica HP-5MS column (30 m x 0.25 mm ID x 0.25 µm film) was used with helium as the carrier gas at a constant flow rate of 1.5 mL/min. The oven temperature was programmed to 50°C for 1 min, increased to 200°C at a rate of 50°C/min, further increased to 320°C at a rate of 3°C/min and then finally held for 25 min. The mass spectrometer detector was set at 70 eV with the transfer line and the quadrupole held at 320°C and 150°C, respectively. The Kovats retention index (KI) (Kovats 1965) was calculated for each hydrocarbon peak after measuring the elution times of the alkane standards run under similar conditions. The hydrocarbon peak areas were calculated for each chromatogram (HP Chem Station, Hewlett Packard) and expressed as a percentage of the total peak area. Shorthand nomenclature is used in the text and Supplementary data to identify the hydrocarbons. CXX represents the total number of carbons in the straight chain with linear alkanes denoted by *n*-CXX. The location of methyl branches was described as x-me for monomethyl alkanes, x,x-dime for dimethyl alkanes, x,x,x-trime for trimethyl alkanes and x,x,x,x-tetrame for tetramethyl alkanes. Interpretation of the mass spectra was performed as described previously (Juárez et al. 2001, Calderón-Fernández et al. 2011).

Statistical analyses - Differences in the relative amounts of cuticle hydrocarbons between species were tested by ANOVA. Statistical significance between means was assessed by Tukey's test ( $\alpha = 0.05$ ) using the SPSS v11.0 software (SPSS, Chicago, IL, USA).

### RESULTS

About 100 components from 21-43 atoms in the carbon backbone were detected in the cuticular hydrocarbons of the *T. sordida* subcomplex (Fig. 1A-E, Supplementary data). Homologous series of *n*-alkanes and single-component and isomeric mixtures of meth-

yl-branched alkanes with one-four methyl substitutions were found in the carbon backbone. Trace levels of *n*alkenes were detected (data not shown) with molecular ions two mass units less than those of the corresponding *n*-alkane. Neither qualitative nor quantitative significant differences were observed between males and females in the populations examined (data not shown). Hydrocarbon component identification together with the corresponding chromatographic retention indices and characteristic mass spectral ions are shown in Supplementary data.

*n*-alkanes formed a continuous series from C21-C33. The proportion of *n*-alkanes varied from 33-45% of the total hydrocarbon fraction within the subcomplex (Supplementary data). Odd-numbered chains prevailed over even-numbered ones, with *n*-C29 predominating. The *n*-C31 chain was the second major component in *T. sor-dida*, *T. guasayana* and *T. patagonica*. In contrast, *n*-C27 was the second major component in *T. garciabesi*, showing a significantly lower amount in *T. guasayana* and *T. patagonica* (1.86 ± 0.27% and 1.77 ± 0.15% respectively, p < 0.05) (Figs 1A-E, 2, Supplementary data).

The methyl-branched fraction was represented by a variety of mono, di, tri and tetramethylalkanes, usually present as isomeric mixtures. Most odd-numbered carbon backbones had the methyl groups inserted at oddnumbered carbon atoms, whereas even-numbered backbones had the methyl groups inserted at both odd and even-numbered carbons.

Internally-branched monomethylalkanes were found as a homologous series of minor components that eluted as isomeric mixtures with both odd and even-numbered carbon skeletons ranging from C25-C43 (Supplementary data). As expected, these isomeric mixtures eluted ahead of the n-alkanes with the same number of total carbons (~73 KI units less); the subterminally-branched components (6 and 7-me) eluted slightly closer to the corresponding straight chains (~60 KI units less). Terminally-branched monomethylalkanes (at branching positions C2-C5) were detected in all species of the subcomplex with alkyl chains ranging from 24-35 carbon atoms. Anteiso branching (3-me) was detected in odd and evennumbered alkyl chains, with the former prevailing as a continuous series ranging from C25-C35. A species-specific pattern was observed in the odd-numbered 3-methylalkanes from C27-C35 (Fig. 3, Supplementary data). In both T. guasayana and T. patagonica, 3-methylalkanes showed increasing proportions from  $0.30 \pm 0.07\%$  and  $0.16\pm0.03\%$  for the 3-me C27 to  $6.08\pm0.24\%$  and  $6.15\pm$ 0.37% for the 3-me C33, respectively. In contrast, the 3-me C29 was the major component ( $6.70 \pm 0.20\%$ ) followed by 3-me C31  $(1.50 \pm 0.20\%)$  in T. sordida, whereas both components exhibited almost similar relative amounts in T. garciabesi (5.09  $\pm$  0.30% and 6.05  $\pm$  0.36% respectively). Small amounts of 3-me isomers of even-numbered chains were also found in some of the species studied. Also, 2, 4 and 5-methylalkanes were scarcely represented, except for 2-me C26 in T. garciabesi (1.83  $\pm$  0.07%) and a series of 4-me components in T. sordida from Rondonópolis (4-me C28, 4-me C30 and 4-me C34 accounting for 1.23)  $\pm 0.04\%$ , 1.74  $\pm 0.06\%$  and 1.92  $\pm 0.15\%$  of the total hydrocarbons, respectively) (Supplementary data).



Fig 1: total ion chromatograms of male specimens of species and populations of the *Triatoma sordida* subcomplex. A: *Triatoma sordida* from Rondonópolis, Brazil; B: *Triatoma sordida* from Córdoba, Argentina; C: *Triatoma garciabesi*; D: *Triatoma guasayana*; E: *Triatoma patagonica*. Peak numbers correspond to peaks listed in Supplementary data.

Dimethylalkanes with both odd and even-numbered carbon chains were detected with the first methyl branch positioned at either terminal (3,x-, 4,x- and 5,x-) or internal carbons. Terminal dimethylalkanes mostly had 7 or more methylene groups separating methyl branches; 3 methylene groups usually separated the methyl groups in internally-branched dimethylalkanes (Supplementary data). The relative amounts of several dimethylalkanes showed significant differences among species (Fig. 4, Supplementary data). For example, the 13,23- and 11,21dime C35 isomers (KI 3562) represented  $2.07 \pm 0.15\%$ and  $1.12 \pm 0.07\%$  of the total hydrocarbons in the T. sordida specimens (Rondonópolis and Córdoba respectively), decreased to  $0.38 \pm 0.03\%$  in *T. garciabesi* and was undetectable in both T. guasayana and T. patagonica. Similarly, the relative amounts of the dimethyl isomers of C37 (KI 3757) varied from  $2.95 \pm 0.13\%$  in *T. sordida* (Rondonópolis) and  $2.12 \pm 0.08\%$  in T. sordida (Córdoba) to  $1.09 \pm 0.07\%$  in *T. garciabesi* and further diminished to  $0.25 \pm 0.06\%$  and  $0.21 \pm 0.06\%$  in *T. guasayana* and T. patagonica, respectively. Conversely, the latter two species had significantly higher relative amounts of the 5,x-dime isomers of C37 and C39 (KI 3782 and 3982 respectively), whereas these components were undetectable or present in trace amounts in T. sordida and T. garcia*besi*. A series of 3,x-dime isomers (x = 7, 9, 11, 13, 15 and17) were detected in odd-numbered carbon chains from C27-C39 (Supplementary data). The 3,x-dime isomers of C37 (KI 3805) and C39 (KI 4005) were the predominant components, eluting with a KI of 100-95 units less than *n*-alkanes with the same total number of carbon atoms (Supplementary data). 3,x-dime isomers of C39 were significantly different among species, particularly between T. sordida populations (Fig. 4). Dimethylalkanes with even-numbered carbon skeletons were represented by

terminal and internal components of 30-40 carbon atoms in the alkyl chain. Terminal components consisted of low amounts of 4,8, 4,16 and 4,18-dime isomers detected in carbon chains C30-C40; only the 4,x-dime isomers of C38 were found in the four species studied, ranging from  $0.42 \pm 0.02\%$  to  $1.34 \pm 0.09\%$  of the total hydrocarbons (Supplementary data). Dimethylalkanes with both methyl branches positioned internally included the 6,10, 12,22, 14,18 and 16,20 isomers with straight backbones from 34-40 carbons (Supplementary data). 6,10-dimethylalkanes co-eluted with the trimethyl components with the same number of carbons in the alkyl chain (Supplementary data), but they were identified by the presence of a strong ion doublet at m/z 98/99 together with a significant ion at m/z 169 corresponding to the presence of the methyl groups at carbons 6 and 10, respectively.

Trimethylalkanes with both odd and even-numbered carbon skeletons from 29-41 carbons were found throughout the subcomplex (Supplementary data). Internallybranched trimethyl alkanes usually had three methylene groups between the first and the second methyl branches and five methylene groups between the second and the third branch (3+5 pattern); to a lesser extent, a 3+3 pattern was also detected. Thus, mixtures of 11,15,21, 11,17, 21, 13,17,21, 13,17,23 and 13,19,23 components prevailed in odd-numbered carbon backbones from C31-C41, while 12,16,22, 12,18,22, 14,18,24 and 14,18,22 components were found in even-numbered backbones from C34-C40. Even-numbered chains with methyl branches positioned at odd carbons were also detected, such as 11,17,21-trime C32, 11,15,21-trime C34 and 13,17,23 plus 13,19,23-trime C36. Internal trimethyl isomers of C37 and C39 were the major methyl-branched components of the subcom-



Fig 2: variation of major *n*-alkanes among species and populations of the *Triatoma sordida* subcomplex. 1: *T. sordida* (Rondonópolis, Brazil); 2: *T. sordida* (Córdoba, Argentina); 3: *Triatoma garciabesi*; 4: *Triatoma guasayana*; 5: *Triatoma patagonica*. Different letters in hydrocarbon amount indicate significant differences (p < 0.05) among groups by Tukey's test. Kovats index (KI) 2700: *n*-C27; KI 2900: *n*-C29; KI 3100: *n*-C31; KI 3300: *n*-C33.

Fig 3: variation of odd-numbered 3-methylalkanes among species and populations of the *Triatoma sordida* subcomplex. 1: *T. sordida* (Rondonópolis, Brazil); 2: *T. sordida* (Córdoba, Argentina); 3: *Triatoma garciabesi*; 4: *Triatoma guasayana*; 5: *Triatoma patagonica*. Different letters in hydrocarbon amount indicate significant differences (p < 0.05) among groups by Tukey's test. Kovats index (KI) 2775: 3-me C27; KI 2975: 3-me C29; KI 3175: 3-me C31; KI 3375: 3-me C33; KI 3575: 3-me C35.

plex, showing a specific pattern of variation. The relative amounts of internal trimethyl isomers of C37 (KI 3780) decreased from  $10.00 \pm 0.60\%$  in T. sordida (Rondonópolis) to 2.66  $\pm$  0.15% in *T. patagonica*, showing the most significant difference between the two T. sordida populations, as well as between T. guasavana and T. patagonica (Fig. 5, Supplementary data). In contrast, the relative amounts of internal trimethyl isomers of C39 (KI 3980) increased from  $2.48 \pm 0.11\%$  in *T. sordida* (Rondonópolis) to  $10.19 \pm 0.40\%$  in *T. patagonica*, showing significant differences among T. sordida from Brazil, T. sordida from Argentina and T. garciabesi (Fig. 5, Supplementary data). Terminal trimethylalkanes included the 2,6,10 isomers of C34, C36 and C38; 3,7,x- and 3,11,15 isomers of C35, C37 and C39 and traces of 4,8,x isomers of C36 and C38 (Supplementary data). The 2,6,10 isomers co-eluted with the internally-branched odd-numbered monomethylalkanes of C35, C37 and C39, respectively (Supplementary data). 3,x,x-trimethyl branching was detected in odd-numbered alkyl chains from C35-C39. Similarly to 2,x,x-trimethyl alkanes, they co-eluted with or close to the internally-branched monomethylalkanes of C36, C38 and C40, respectively (Supplementary data).

Few tetramethylalkanes were detected in the species subcomplex. A complete series was found in *T. patagonica* with small amounts of 4,8,12,16-tetrame isomers of C34, C36 and C38. Also, the 3,7,11,15-tetrame of C35 (KI 3653) was detected throughout the species subcomplex, peaking at  $1.52 \pm 0.12\%$  in *T. sordida* (Rondonópolis).

## DISCUSSION

Insect cuticular hydrocarbons often occur as complex blends of straight-chain and methyl-branched components, with saturated and unsaturated alkyl chains ranging from about 21 to more than 50 carbons (Howard & Blomquist 2005, Blomquist 2010). Among other functions, cuticle hydrocarbons have been shown to participate in protecting insects against desiccation, microbial attack and insecticide penetration (Juárez 1994, Napolitano & Juárez 1997, Pedrini et al. 2009) and also play a key role in chemical communication as species, mate and nestmate recognition cues (Blomquist & Bagnères 2010). In triatomines and several other insect groups, the hydrocarbon pattern is mainly under genetic control (Juárez & Calderón-Fernández 2007, Wicker-Thomas & Chertemps 2010). Thus, the expression of specific cuticular hydrocarbons reflects the expression of genes coding for hydrocarbon biosynthetic enzymes. Accordingly, triatomine cuticle hydrocarbons have been used as characters to address taxonomic analysis either at genera, species and populations (Juárez et al. 2000, Calderón-Fernández et al. 2005, 2011, 2012).

Complex mixtures of hydrocarbons were detected in the four species of the T. sordida subcomplex. These mixtures included straight chain alkanes and a variety of mono, di, tri and tetramethylalkanes comprising alkyl chains from 21-43 carbons. In triatomines, an abundant number of tetramethylalkanes were found in *R. prolixus* (Juárez et al. 2001) and T. dimidiata (Calderón-Fernández et al. 2011). However, they were not detected in T. infestans, a phylogenetically similar species (Juárez & Blomquist 1993). Similar to their morphological features, the species of this subcomplex shared a similar hydrocarbon pattern that can be distinguished from the hydrocarbon pattern found in other triatomine species. Therefore, mostly quantitative differences in the cuticular hydrocarbon composition were detected, together with some minor qualitative differences. Several hydrocar-



Fig 4: variation of some dimethylalkanes among species and populations of the *Triatoma sordida* subcomplex. 1: *T. sordida* (Rondonópolis, Brazil); 2: *T. sordida* (Córdoba, Argentina); 3: *Triatoma garciabesi*; 4: *Triatoma guasayana*; 5: *Triatoma patagonica*. Different letters in hydrocarbon amount indicate significant differences (p < 0.05) among groups by Tukey's test. Kovats index (KI) 3562: 11,x and 13,xdime C35; KI 3757: 11,x, 13,x and 15,x-dime C37; KI 3782: 5,x-dime C37; KI 3982: 5,x-dime C39; KI 4005: 3,x-dime C39.



Fig 5: variation of some trimethylalkanes among species and populations of the *Triatoma sordida* subcomplex. 1: *T. sordida* (Rondonópolis, Brazil); 2: *T. sordida* (Córdoba, Argentina); 3: *Triatoma garciabesi*; 4: *Triatoma guasayana*; 5: *Triatoma patagonica*. Different letters in hydrocarbon amount indicate significant differences (p < 0.05) among groups by Tukey's test. Kovats index (KI) 3581: 11,x,x and 13,x,x-trime C35; KI 3780: 13,x,x and 15,x,x-trime C37; KI 3980: 13,x,x and 15,x,x-trime C40.

bons showed patterns of variation that helped to clearly discriminate the species and populations analysed in this work. T. guasayana and T. patagonica showed simpler cuticular hydrocarbon profiles with smaller amounts (or absence) of several mono, di and trimethyl alkanes from C25-C35 (Fig. 1D, E, Supplementary data). In contrast, they showed relatively larger amounts of some terminally-branched components (3-me C33, 3-me C35 and 5,x-dime isomers of C37 and C39) that were present in minor amounts or absent in T. garciabesi and T. sordida. T. garciabesi showed a more complex chromatographic profile, especially in terms of the number of hydrocarbon peaks (Supplementary data). However, the profile is clearly similar to that of T. sordida, regardless of its origin (Fig. 1A-C). T. garciabesi showed no evident qualitative differences, but significant quantitative differences in several hydrocarbons, such as 2-me C28, 3-me C27, 3-me C29 etc (Supplementary data). The analysis of the two geographically distant T. sordida populations revealed that they had significant quantitative differences in several components such as the 4-me series from C28-C34, the 4,x-dime series from C30-C34, the 3-me C29 and C31, the trimethyl isomers of C35, C37 and C39 and others (Figs 3-5, Supplementary data). Some of these differences were equivalent to those found between T. garciabesi and T. sordida, thus suggesting that T. sordida is too heterogeneous to be considered a single species. Interestingly, previous analyses using a variety of techniques suggested that T. sordida consisted of at least two cryptic species (Panzera et al. 1997, Noireau et al. 1998).

After the classification proposed by Lent and Wygodzinsky (1979), the taxonomic status of the T. sordida species subcomplex has been scarcely studied, probably because these species were considered secondary vectors without relevance in transmitting Chagas disease. Furthermore, the few studies performed using morphological, morphometric or chromosomal characters have led to distinct and sometimes contradictory results. T. garciabesi was synonymised with T. sordida some years after its description (Lent & Wygodzinsky 1979); later on, it was revalidated as a valid species (Jurberg et al. 1998). The analysis of several metric variables of head and male genitalia completely differentiated T. sordida from both T. guasayana and T. patagonica, but failed to differentiate the latter two species (Gorla et al. 1993). In contrast, cytogenetic and enzyme electrophoresis studies clearly differentiated T. guasayana from both T. patagonica and T. sordida. These analyses also separated T. sordida into two groups (Brazil and Argentina), but a scarce differentiation was achieved between T. sordida populations from the wet and dry regions of Argentina (those of the dry region were formerly described as T. garciabesi) (Panzera et al. 1997). More recently, an extensive phylogenetic analysis of the Triatominae subfamily based on molecular data placed the species into a different triatomine group (Hypsa et al. 2002). Thus, current evidence suggests that the taxonomy of the subcomplex should be revised, especially regarding the relationship between T. guasavana and T. patagonica, as well as the intraspecific variation of the so-called "T. sordida." The use of cuticular hydrocarbons as taxonomic characters

might help to elucidate the precise relationship between these species, as well as their intraspecific variability. This knowledge would provide valuable information to Chagas disease control programs in their efforts to monitor and control triatomine populations.

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	Identification by gas chromatograph	ity coupled to mass spectrometry of the cuticular nyarocarbons found in the species of t	ne <i>triatoma sorataa</i> subcomplex
Peak <sup>a</sup>	KI	Hydrocarbon	Diagnostic ions
	2100	<i>n</i> -C21	296
7	2200	n-C22	310
3	2300	n-C23	324
4	2400	n-C24	338
5	2461	2-me C24	309, 337
9	2500	n-C25	352
7	2527	11- + 13-me C25	168/169, 196/197, 224/225
8	2600	<i>n</i> -C26	366
6	2627	12-+13-+11-+10-me C26	182/183, 224/225; 196/197, 210/211:
			168/169, 238/239; 154/155, 252/253
10	2660	2-me C26	336/337 365
11	2700	<i>n</i> -C27	380
12	2727	13- + 11-me C27	196/197, 224/225; 168/169, 252/253
13	2775	3-me C27	56/57, 365
14	2800	n-C28	394
14'	2804	3,17- + 3,15-dime C27	56/57; 267, 168/169; 239, 196/197; 379
15	2828	13-+12-+11-+14-me C28	196/197, 238/239; 182/183, 252/253; 168/169, 266/267; 210/211, 224/225
16	2857	4-me C28	70/71, 336/337, 365; 365, 393
17	2875	3-me C28	56/57, 379
18	2900	и-С29	408
19	2927	11-+13-+15-+9-me C29	168/169, 280/281; 196/197, 252/253; 224/225; 140/141, 308/309
20	2939	7-me C29	112/113, 336/337
21	2950	5-me C29	84/85, 336/337, 365

Supplementary data

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Peak <sup>a</sup>	KI	Hydrocarbon	Diagnostic ions
22	2958	13,17- + 11,15-dime C29	196/197, 267; 168/169, 239, 224/225, 295
22`	2961	9,13-dime C29	140/141, 211, 252/253, 323
23	2975	3-me C29	56/57, 364/365, 393
24	2993	7,13,17-trime C29	112/113, 211, 281, 196/197, 267, 365
25	3000	<i>n</i> -C30	422
25'	3004	3,17- + 3,15-dime C29	57; 267, 196/197; 239, 224/225; 407
26	3028	13- + 14- + 15-me C30	196/197, 266/267; 210/211, 252/253; 224/225, 238/239
27	3044	6-me C30	98/99, 336/337, 365
28	3058	4-me C30	70/71, 364/365, 392/393
29	3062	2-me C30	393, 421
30	3073	3-me C30	56/57, 379, 407
31	3092	4,16-+4,18-dime C30	
			233, 224/223, 281, 196/197; 407
32	3095	7,13,17-trime C30	112/113, 211, 281, 210/211, 281, 379
33	3100	<i>n</i> -C31	436
34	3129	13- + 11- + 15- + 9-me C31	196/197, 280/281; 168/169, 308/309; 224/225, 252/253; 140/141, 336/337
35	3140	7-me C31	112/113, 364/365
36	3157	13,17- + 11,15-dime C31	196/197, 267, 224/225, 295; 168/169, 239, 252/253, 323
37	3163	11,21-dime C31	168/169, 323
38	3175	3-me C31	56/57, 392/393 421
39	3183 9,1	13,19-+ 11,15,19-+ 11,15,21-trime C31	140/141, 211, 309, 196/197, 295 365; 168/169, 239, 309, 196/197, 267, 337; 168/169, 239, 267, 309, 337
40	3195	7,13,17-trime C31	112/113, 211, 281, 224/225, 295, 393
41	3202	n-C32	450

Peak <sup>a</sup>	KI	Hydrocarbon	Diagnostic ions
41'	3204	3,15- + 3,17-dime C31	56/57; 239, 252/253; 267, 224/225; 435
42	3208	3,7-dime C31	56/57, 127, 364/365, 435
43	3223	7,11,17,21-tetrame C31	112/113, 183, 281, 351, 168/169, 239, 337, 407
44	3236	8-me C32	126/127, 364/365
45	3258	4-me C32	70/71, 392/393, 421
46	3275	3-me C32	56/57, 407, 435
47	3282	11,17,21-trime C32	168/169, 267, 337, 182/183, 253, 351
48	3290	4,8-dime C32	70/71, 141, 364/365, 435
49	3300	n-C33	464
50	3327	11-+13-me C33	168/169, 336/337; 196/197, 308/309
51	3362	11,21-+15,19-dime C33	168/169, 323, 196/197, 351; 224/225, 295
52	3375	3-me C33	56/57, 420/421, 449
53	3383	11,15,21-+ 11,17,21-trime C33	168/169, 196/197, 239, 267, 295, 337, 365
54	3405	3,17- + 3,15-dime C33	56/57; 267, 252/253; 239, 280/281; 463
55	3408	3,7- + 3,9-dime C33	56/57; 127, 392/393; 155, 364/365; 463
56	3428	12-me C34	182/183, 336/337
57	3457	4-me C34	70/71, 448/449
		12,22-dime C34	182/183, 337, 196/197, 351
58	3473	3-me C34	56/57, 463
59	3479	12,16,22-+12,18,22-+11,15,21-trime C34	182/183, 196/197, 253, 267, 295, 351, 365; 168/169, 239, 337, 210/211, 309, 379
60	3490	4,16- + 4,18-dime C34	70/71; 253, 280/281; 281, 252/253; 463
61	3492	4,8-dime C34	70/71, 141, 392/393, 463

	KIHydrocarbon529 $2,6,10$ -trime C34540 $2,6,10$ -trime C34541 $3,12,16$ -tetrame C34562 $4,8,12,16$ -tetrame C34563 $3,17,21$ + 11,21-dime C35573 $13,17,23$ + 13,17,21+ 11,15,21-trime C35581 $3,17,23$ + 13,17,21+ 11,15,21-trime C35605 $3,15$ + 5,13+ 5,9+ 5,11-dime C35605 $3,15$ + 5,13- + 3,11-dime C35606 $3,11,15$ - 4,11- 11- 11- 11-me C35617 $3,71,11$ -trime C35626 $12$ - 14, + 11- + 13-me C3663 $3,71,11$ -trime C3563 $3,71,11$ -trime C3564 $3,71,11,15$ -trime C35658 $3,71,11,15$ -trime C35678 $13,17,23$ - 112,16,22- + 12,18,22- + 11,15,23-trime	Diagnostic ions   113, 183, 364/365, 435, 505   113, 183, 364/365, 135, 1421, 491.   196/197, 336/337   70/71, 141, 211, 281, 280/281, 351, 421, 491.   196/197, 351, 196/197, 351, 1241, 491.   196/197, 267, 295, 337, 355;   196/197, 267, 295, 337, 355;   196/197, 267, 295, 337, 355;   196/197, 267, 295, 337, 355;   196/197, 267, 295, 337, 355;   196/197, 267, 295, 337, 324/225, 295, 365;   198/169, 239, 337, 224/225, 233, 393   84/85;   267, 280/281;   196/197, 267, 295, 365;   196/197, 267, 337, 224/225, 233, 393   196/197, 267, 308/309;   267, 280, 281;   36/37;   113, 336/337;   182/183, 364/365;   283, 364/365;   113, 336/337;   113, 336/337;   113, 336/337;   113, 336/337;   182/183, 364/365;   282, 33, 364/365;   283, 364/365;   113, 336/337;   113, 336/337;   113, 336/337;   113, 336/337;   113, 324/365, 443, 519   98/99, 169, 392/393, 463   19
3690	4,16- + 4,14-dime C36	10%, 10%, 20%, 20%, 20%, 20%, 20%, 20%, 20%, 2
3727	2,6,10-trime C36 11- + 13- + 15-me C37	43, 113, 183, 392/393, 463, 533 168/169, 392/393; 196/197, 364/365; 224/225, 336/337

Peak <sup>a</sup>	KI	Hydrocarbon	Diagnostic ions
85	3957	15,19- + 17,21- + 13,17- + 13,23-dime C39	224/225, 295, 308/309, 379; 252/253, 323, 280/281, 351; 196/197, 267, 336/337, 407; 196/197, 351, 252/253, 407
86	3980	13,17,21- + 13,17,23 + 15,19,23-trime C39	196/197, 267, 337, 280/281, 351, 421; 196/197, 267, 365, 252/253, 351, 421; 224/225, 295, 365, 252/253, 323, 393
86'	3982	5,15- + 5,17-dime C39	84/85; 239, 364/365; 267, 336/337; 519
87	4005	3,15- + 3,17-dime C39	57; 239, 364/365; 267, 336/337; 547
88	4026	12-+14-+16-+18-me C40	182/183, 420/421; 210/211, 392/393; 238/239, 364/365; 266/267, 336/337
89	4032	3,7,11- + 3,7,15- + 3,7,17-trime C39	57, 127; 197, 420/421; 253, 364/365; 281, 336/337; 491, 561
06	4060	16,20- + 18,22-dime C40	238/239, 309, 308/309, 379; 266/267, 337, 280/281, 351
91	4080	14,18,24-+14,18,22-+16,20,24-trime C40	210/211, 281, 379, 252/253, 351, 421; 210/211, 281, 351, 280/281, 351, 421; 238/239, 309, 379, 252/253, 323, 393
92	4127	13-+11-+15-+17-me C41	196/197, 420/421; 168/169, 448/449; 224/225, 392/393; 252/253, 364/365
93	4157	17,21- + 19,23-dime C41	252/253, 323, 308/309, 379; 280/281, 351
94	4179	13,17,21-+15,19,23-+17,21,25-trime C41	196/197, 267, 337, 308/309, 379, 449; 224/225, 295, 365, 280/281, 351, 421; 252/253, 323, 393
95	4327	5,15-dime C41 17- + 19-me C43	85, 239, 392/393, 547 252/253, 392/393; 280/281, 364/365

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		Cuticular hydrocarbon composition	of species of the Iriatoma	sorataa subcomp	lex		
Peak <sup>a</sup>	KI	Hydrocarbon	<i>T. sordida</i> Rondonópolis	<i>T. sordida</i> Córdoba	Triatoma garciabesi	Triatoma guasayana	Triatoma patagonica
_	2100	<i>n</i> -C21	tr	$0.15 \pm 0.09$	tr	tt	 
2	2200		t t	$0.19 \pm 0.03$	t t	t t	L H
ŝ	2300	n-C23	$0.11 \pm 0.01$	$0.24\pm0.01$	$0.15 \pm 0.04$	tr	tr
4	2400	n-C24	tr	$0.21 \pm 0.07$	$0.20 \pm 0.02$	tr	tr
5	2461	2-me C24	·	tr	$0.12 \pm 0.07$	ı	ı
9	2500	<i>n</i> -C25	$0.32 \pm 0.11$	$0.54\pm0.20$	$0.83\pm0.18$	$0.10\pm0.06$	tr
7	2527	11-+13-me C25	$0.18\pm0.02$	$0.10\pm0.02$	$0.11 \pm 0.03$		·
8	2600	n-C26	$0.14 \pm 0.03$	$0.22 \pm 0.03$	$0.71 \pm 0.11$	tr	tr
6	2627	12-+13-+11-+10-me C26	$0.17\pm0.03$	$0.13 \pm 0.02$	$0.11 \pm 0.03$	ı	ı
10	2660	2-me C26	$0.10\pm0.03$	$0.12 \pm 0.03$	$1.83 \pm 0.07$	$0.19\pm0.03$	tr
11	2700	n-C27	$4.13 \pm 0.39$	$6.43\pm0.39$	$5.13 \pm 0.85$	$1.86 \pm 0.27$	$1.77 \pm 0.15$
12	2727	13-+11-me C27	$0.92\pm0.02$	$0.94\pm0.02$	$0.63 \pm 0.03$	$0.20 \pm 0.03$	$0.19\pm0.03$
13	2775	3-me C27	$0.70\pm0.10$	$0.73 \pm 0.07$	$1.51 \pm 0.30$	$0.30 \pm 0.07$	$0.16\pm0.03$
14	2800	n-C28	$2.25\pm0.17$	$2.51\pm0.17$	$2.19 \pm 0.19$	$1.13 \pm 0.14$	$1.41 \pm 0.10$
14`	2804	3,17- + 3,15-dime C27	tr	tr	$0.24\pm0.03$	·	ı
15	2828	13-+12-+11-+14-me C28	$0.17 \pm 0.01$	$0.17 \pm 0.01$	$0.16\pm0.02$	tr	tr
16	2857	4-me C28	$1.23 \pm 0.04$	$0.37\pm0.04$	$0.77 \pm 0.04$	$0.17 \pm 0.02$	$0.10 \pm 0.02$
17	2875	3-me C28	$0.14\pm0.02$	$0.26\pm0.03$	$0.27\pm0.02$		
18	2900	<i>n</i> -C29	$20.38 \pm 0.66$	$25.53 \pm 0.53$	$18.00\pm0.62$	$22.92 \pm 0.91$	$29.26 \pm 1.59$
19	2927	11-+13-+15-+9-me C29	$0.57\pm0.04$	$0.48\pm0.03$	$0.39\pm0.03$	$0.33 \pm 0.02$	$0.48\pm0.03$
20	2939	7-me C29	$0.14\pm0.01$	$0.21\pm0.02$	$0.18 \pm 0.02$	ı	ı
21	2950	5-me C29	$0.13\pm0.01$	$0.41 \pm 0.02$	$0.43 \pm 0.03$	tr	tr
22	2958	13,17- + 11,15-dime C29	$0.19\pm0.03$	$0.26\pm0.05$	$0.28\pm0.03$	$0.18\pm0.01$	
22'	2961	9,13-dime C29	$0.27\pm0.03$	ı	,		
23	2975	3-me C29	$1.77\pm0.18$	$6.70 \pm 0.20$	$5.09 \pm 0.30$	$0.98 \pm 0.12$	$0.15\pm0.03$
24	2993	7,13,17-trime C29	$0.21\pm0.03$	ı	$0.69\pm0.07$	ı	ı
25	3000	n-C30	$0.52\pm0.03$	$0.87\pm0.04$	$0.66\pm0.03$	$0.76\pm0.04$	$0.71 \pm 0.03$
25'	3004	3,17- + 3,15-dime C29	$0.66\pm0.02$	ı	ı	ı	ı
26	3028	13 - 14 - 15 - me C30	$0.16\pm0.01$	tr	tr	tr	tr
27	3044	6-me C30	$0.27 \pm 0.01$	ı	$0.16\pm0.01$	ı	tr
28	3058	4-me C30	$1.74\pm0.06$	$0.29\pm0.03$	$0.33\pm0.03$	tr	tr
29	3062	2-me C30	tr	tr	$0.13 \pm 0.01$	tr	·
30	3073	3-me C30		tr	$0.29\pm0.01$	$0.20 \pm 0.01$	tr
31	3092	4,16- + 4,18-dime C30	$0.64\pm0.02$	tr	·		·
32	3095	7,13,17-trime C30		ı	$0.12 \pm 0.01$	ı	ı
33	3100	n-C31	$7.66 \pm 0.26$	$6.47 \pm 0.25$	$3.74\pm0.30$	$12.85 \pm 0.37$	$6.24\pm0.21$
34	3129	13-+11-+15-+9-me C31	$0.27 \pm 0.02$	$0.18 \pm 0.02$	tr	$0.17 \pm 0.02$	$0.10 \pm 0.08$
							<b>≜</b>

na nica		ı		$\pm 0.16$	ı	ı	$\pm 0.02$	ı		ı		$\pm 0.02$	$\pm 0.02$			$\pm 0.20$	ı	$\pm 0.02$	$\pm 0.37$	ı	$\pm 0.03$	tr		$\pm 0.02$	+ 0.01	$\pm 0.03$		tr	$\pm 0.03$		$\pm 0.03$	ı	$\pm 0.04$	$\pm 0.05$		$\pm 0.04$	tr		tr	1
Triato. patago				2.43			0.29					0.26	0.30			2.19		0.13	6.15		0.33			0.15	010	0.18			0.12		0.20		0.56	0.72		0.79				
Triatoma guasayana	ı	$0.27\pm0.06$	I	$5.36 \pm 0.75$	I	I	$0.60\pm0.02$	ı	ı	ı	ı	$0.33 \pm 0.02$	$0.37 \pm 0.01$	I	I	$4.68\pm0.18$	ı	$0.20\pm0.08$	$6.08 \pm 0.24$	I	$0.23 \pm 0.02$	$0.10 \pm 0.02$	I	$0.13 \pm 0.02$	$0.10 \pm 0.02$		ı	tr	$0.24\pm0.03$		ı	ı	$0.80\pm0.08$	$0.63\pm0.06$		$0.45 \pm 0.04$	tr		tr	
Triatoma garciabesi	$0.12 \pm 0.01$	$0.13\pm0.03$		$6.05 \pm 0.36$	$0.33\pm0.03$	$0.33 \pm 0.02$	tr	$0.43\pm0.02$	$0.75\pm0.05$	$0.21 \pm 0.02$	·	$0.44\pm0.03$	$0.14\pm0.02$	$0.21 \pm 0.03$	tr	$1.38 \pm 0.19$	tr	$0.16\pm0.02$	$0.91 \pm 0.07$	$0.18 \pm 0.10$	$0.18\pm0.02$	$0.21 \pm 0.02$	tr	$0.20 \pm 0.02$	Ţ	t t	$0.10 \pm 0.01$	tr	$0.31\pm0.02$		ı	$0.38\pm0.03$	·	$0.82 \pm 0.04$		$0.21 \pm 0.02$	$0.21 \pm 0.02$		tr	
<i>T. sordida</i> Córdoba	$0.10 \pm 0.01$	$0.21 \pm 0.02$		$1.50 \pm 0.20$	$0.11 \pm 0.09$	$0.10 \pm 0.01$	tr	$0.39\pm0.03$	$0.28\pm0.02$	·		$0.11 \pm 0.01$	ı		tr	$1.25\pm0.10$	$0.10 \pm 0.01$	$0.64\pm0.05$	$0.49\pm0.05$	$0.21 \pm 0.15$	$0.22\pm0.03$	$0.14 \pm 0.01$	tr	$0.48 \pm 0.07$	I	$0.14 \pm 0.01$		·	$0.64\pm0.03$			$1.12 \pm 0.07$		$1.65\pm0.04$		$0.33 \pm 0.04$	$0.24 \pm 0.02$		tr	
<i>T. sordida</i> Rondonópolis	$0.29 \pm 0.02$	$0.21 \pm 0.02$	$0.34 \pm 0.02$	$0.42 \pm 0.06$	$0.30 \pm 0.11$	$0.22 \pm 0.02$	ı	$0.98\pm0.05$	tr	ı	$0.22 \pm 0.02$	$0.59\pm0.05$	I	ı	$0.68 \pm 0.04$	$2.57 \pm 0.17$	$0.43 \pm 0.02$	$1.49\pm0.12$	$0.39\pm0.05$	$0.80 \pm 0.50$	$0.54\pm0.03$	ı	$0.23 \pm 0.12$	$1.92 \pm 0.15$	I	$0.53 \pm 0.02$		$0.66\pm0.02$	$0.64\pm0.03$		$0.34\pm0.03$	$2.07 \pm 0.15$	ı	$3.64\pm0.10$		$0.61 \pm 0.08$	$0.51 \pm 0.02$		tr	
	7-me C31	13,17- + 11,15-dime C31	11,21-dime C31	3-me C31	9,13,19-+11,15,19-+11,15,21-trime C31	7,13,17-trime C31	n-C32	3,17- + 3,15-dime C31	3,7-dime C31	7,11,17,21-tetrame C31	8-me C32	4-me C32	3-me C32	11,17,21-trimethyl C32	4,8-dime C32	n-C33	11-+13-me C33	11,21- + 15,19-dime C33	3-me C33	11,15,21- + 11,17,21-trime C33	3,17- + 3,15-dime C33	3,7- + 3,9-dime C33	12-me C34	4-me C34	12,22-dime C34 3 ma C34	12 16 22- + 12 18 22- + 11 15 21-trime C34	4.16-+4.18-dime C34	4.8-dime C34	2,6,10-trime C34	11-+13-me C35	4,8,12,16-tetrame C34	13,23- + 11,21-dime C35	3-me C35	13,17,23 - + 13,17,21 - + 11,15,21-trime C35	5,17- + 5,13- + 5,9- + 5,11-dime C35	3,15-+3,13-+3,11-dime C35	3,11,15-trime C35	12 - + 14 - + 11 - + 13 - me C36	3,/,11-trime C35	
Hydrocarbon																																								
a KI	3140	3157	3163	3175	3183	3195	3202	3204	3208	3223	3236	3258	3275	3282	3290	3300	3327	3362	3375	3383	3405	3408	3428	3457	2772	3479	3490	3492	3529		3540	3562	3575	3581		3605	3626		3034	
Peak	35	36	37	38	39	40	41	41'	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62		63	64	65	99		67	68	ç	69	i

Peaka	KI	Hydrocarbon	<i>T. sordida</i> Rondonópolis	<i>T. sordida</i> Córdoba	Triatoma garciabesi	Triatoma guasayana	Triatoma patagonica
70 71	3653 3678	3,7,11,15-tetrame C35 6,10-dime C36	$1.52 \pm 0.12$ $1.95 \pm 0.06$	$0.62 \pm 0.07$ $0.62 \pm 0.02$	$0.47 \pm 0.05$ $0.59 \pm 0.07$	$0.25 \pm 0.02$ 1.19 $\pm 0.06$	tr 1.25 ± 0.04
CL	0098	13,17,23-+13,19,23-+12,16,22-+12,18,22-+14,18,22-+11,15,23-trime C36	$0.14 \pm 0.01$	$0.13 \pm 0.01$	$0.18 \pm 0.02$		$0.19 \pm 0.04$
73	3727	2,6,10-trime C30	$2.20 \pm 0.18$	$1.66 \pm 0.12$	$0.10 \pm 0.02$ $1.07 \pm 0.12$	$0.81 \pm 0.14$	$0.51 \pm 0.04$
		11-+13-+15-me C37					
74	3740	4,8,12,16-tetrame C36	ı	ı	ı	ı	$0.55\pm0.06$
75	3757	13,17-+11,15-+15,19-+11,21-dime C37	$2.95 \pm 0.13$	$2.12 \pm 0.08$	$1.09\pm0.07$	$0.25 \pm 0.06$	$0.21 \pm 0.06$
76	3780	13,17,21-+13,17,23-+13,19,23-+15,19,23-trime C37	$10.00\pm0.60$	$7.13 \pm 0.24$	$6.95 \pm 0.30$	$5.41 \pm 0.20$	$2.66 \pm 0.15$
76,	3782	5,13-+5,15-+5,17-dime C37		tr	tr	$1.35\pm0.09$	$1.78 \pm 0.12$
LL	3805	3,11 - 3,15 - 3,13 - 3,17 - dime C37	$3.00 \pm 0.23$	$2.84\pm0.27$	$3.76 \pm 0.30$	$1.86\pm0.07$	$2.72 \pm 0.20$
78	3827	12- + 14- + 16-me C38	$1.15\pm0.04$	$1.40\pm0.05$	$1.10\pm0.05$	$0.82 \pm 0.04$	$0.75 \pm 0.04$
79	3835	3,7,11- + 3,7,15-trime C37		tr	$0.67\pm0.03$	$0.20\pm0.05$	$0.26\pm0.08$
80	3858	14,18-+12,22-+16,20-dime C38	$1.41 \pm 0.03$	$1.25\pm0.03$	$1.11 \pm 0.03$	$0.42 \pm 0.02$	$0.77 \pm 0.03$
81	3877	6.10-dime C38	$2.56 \pm 0.08$	$2.30 \pm 0.08$	$2.84\pm0.13$	$3.72 \pm 0.14$	$5.65 \pm 0.18$
		14,18,22- + 14,18,24-trime C38					
82	3890	4,16-+4,18-dime C38	$0.46\pm0.02$	$0.76\pm0.03$	$1.34\pm0.09$	$0.42 \pm 0.02$	$0.66\pm0.02$
83	3928	2,6,10-trime C38	$0.55\pm0.03$	$1.40 \pm 0.07$	$1.53\pm0.04$	$1.42 \pm 0.12$	$3.01 \pm 0.07$
		13-+15-+11-+17-me C39					
84	3940	4,8,12,16-tetrame C38					$0.34 \pm 0.02$
85	3957	15,19- + 17,21- + 13,17- + 13,23-dime C39	$1.05\pm0.07$	$2.18\pm0.11$	$1.50\pm0.15$	$1.30 \pm 0.13$	$1.84\pm0.06$
86	3980	13,17,21- + 13,17,23 + 15,19,23-trime C39	$2.48 \pm 0.11$	$6.72 \pm 0.25$	$8.49\pm0.24$	$9.65\pm0.50$	$10.19\pm0.40$
86'	3982	5,15- + 5,17-dime C39		tr	tr	$0.84\pm0.02$	$0.89\pm0.02$
87	4005	3,15- + 3,17-dime C39	$0.41\pm0.03$	$2.43\pm0.08$	$4.05\pm0.04$	$2.78\pm0.06$	$3.87\pm0.07$
88	4026	12- + 14- + 16- + 18-me C40		tr	tr	$0.78\pm0.02$	$0.77 \pm 0.02$
89	4032	3,7,11 - 3,7,15 - 3,7,17-trime C39		tr	$0.47 \pm 0.02$	tr	tr
90	4060	16,20- + 18,22-dime C40		tr	tr	$0.25\pm0.02$	$0.45 \pm 0.02$
91	4080	14,18,24- + 14,18,22- + 16,20,24-trime C40	tr	$0.30\pm0.03$	$0.80\pm0.05$	$0.73\pm0.05$	$1.78\pm0.08$
92	4127	13- + 11- + 15- + 17-me C41		tr	$0.37\pm0.04$	$0.46\pm0.04$	$0.87\pm0.06$
93	4157	17,21- + 19,23-dime C41		tr	tr	$0.15\pm0.03$	$0.31 \pm 0.03$
94	4179	13,17,21- + 15,19,23- + 17,21,25-trime C41		$0.20\pm0.04$	$0.46\pm0.03$	$0.64\pm0.04$	$0.59\pm0.04$
		5,15-dime C41					
95	4327	17- + 19-me C43	·	ı	ı	$0.14 \pm 0.01$	$0.10 \pm 0.01$
<i>a</i> : pe	ak numba ard error	rrs correspond to peaks in Fig. 1; KI: Kovats index; tr: trace amounts (< $0.10\%$ ); of males (no significantly different from females).	; -: not detected. ]	Numbers in colun	nns represent hyd	rocarbon relative	amount (mean% ±